

Beam deflection measurement of bound-electronic and rotational nonlinear refraction in molecular gases

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Abstract: A polarization-resolved beam deflection technique is used to separate the bound-electronic and molecular rotational components of nonlinear refractive transients of molecular gases. Coherent rotational revivals from N₂, O₂, and two isotopologues of carbon disulfide (CS₂), are identified in gaseous mixtures. Dephasing rates, rotational and centrifugal distortion constants of each species are measured. Polarization at the magic angle allows unambiguous measurement of the bound-electronic nonlinear refractive index of air and second hyperpolarizability of CS₂. Agreement between gas and liquid phase second hyperpolarizability measurements is found using the Lorentz-Lorenz local field correction.

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1. Introduction

The nonlinear optical (NLO) response of gases under intense laser fields has been extensively investigated and plays a critical role in applications of pulse compression [1], high harmonic generation [2], attosecond pulse shaping [3, 4], control of filamentation [5–8], and separation of molecular isotopologues [9]. A molecular gas with an anisotropic polarizability exhibits nonlinear refraction (NLR) that originates from the nearly instantaneous response of bound electrons and the delayed response from field induced molecular rotation. Coherent excitation of many rotational states via coherent Raman scattering with a femtosecond pulse results in periodic revivals in the degree of alignment of the molecular ensemble [10–12].

Various time-resolved techniques based on the typical pump-probe geometry have been applied to measure rotational revivals in molecular gases. An intense pump pulse induces rotation of the molecules towards its polarization direction and a variably delayed probe pulse experiences a phase shift from the subsequent refractive index change. The Optical Kerr Effect (OKE) [13] was the earliest method used to measure rotational revivals in gaseous CS₂ [10, 14]. OKE only measures the induced birefringence and lacks the flexibility to measure the polarization dependence of the nonlinear response. Degenerate four wave mixing (DFWM) [15] has also been used to measure low density CS₂ gas, precisely determining the rotational and centrifugal distortion constants [16]. However, without the implementation of more complicated optical-heterodyne-detection (OHD) [17], neither OKE nor DFWM are able to measure the sign of the refractive index change. Interferometric techniques have also been employed [18, 19], which use an additional reference pulse to interfere with the probe and the signal to give both the magnitude and sign of the phase shift. However, such methods may be highly sensitive to environmental perturbations, unless single-shot techniques are employed [18]. Alternatively, in time-resolved Coulomb explosion imaging an intense probe pulse initiates coulomb explosion and the velocities of the ions are angularly resolved to map out the molecular alignment [20, 21]. This technique does not provide information on the bound-electronic nonlinear response.

In this paper, we apply our recently developed beam deflection (BD) technique [22] to characterize the NLR response of ambient air and gas phase CS₂. BD is an excite-probe

technique capable of measuring the absolute magnitude, sign, temporal dynamics, and polarization dependence of NLR with high sensitivity. Measurable induced phase changes as small as 0.3 mrad (optical path length changes of $\lambda/20,000$) have been demonstrated. It is capable of measuring a variety of ultrafast nonlinear optical processes, including nearly instantaneous bound-electronic NLR and two-photon absorption, as well as noninstantaneous effects. We have previously performed absolutely calibrated BD measurements to fully characterize the NLR response of liquid CS₂, unambiguously separating the four major nonlinear responses (bound-electronic plus three nuclear contributions) including their temporal and polarization dependences [23]. In principle, the BD technique is similar to other multi-beam nonlinear measurement techniques, such as OKE, DFWM, and interferometric methods, but has a number of distinct advantages. Unlike the OKE and DFWM, the measured signal in a BD measurement is directly proportional to the induced refractive index change, yielding both the magnitude and sign of the NLR without the requirement of OHD. It is therefore a simpler technique and may be more easily implemented than interferometric methods [19]. BD also allows complete freedom in selecting polarization of both the pump and probe beams, making it more flexible than OKE which measures only the induced birefringence. In gases, BD is also able to measure refractive index revivals over many periods, along with the dephasing rate, enabling determination of the rotational Raman spectrum. Measurement of air demonstrates the BD technique's capability to identify and measure properties of individual constituents of a gaseous mixture. We utilize this in gaseous CS₂, where different vibrational states and isotopologues are measured.

By investigating the polarization dependence, BD also measures the tensor symmetry of the third-order nonlinearity. With a probe polarized at the "magic angle" (54.7°) with respect to that of the excitation polarization, the rotational contribution is completely eliminated from the measurement [24]. This isolates the bound-electronic NLR and allows unambiguous determination of the nonlinear refractive index $n_{2,el}$ and the second hyperpolarizability γ [23]. Gas phase measurements of γ allow direct comparison to theoretical calculations [25, 26] and tests of local field effects in liquid phase measurements [27, 28]. We find agreement between gas phase measurements of γ of CS₂, and our liquid phase measurements [23] when using Lorentz-Lorenz local field correction [28, 29]. The BD technique provides a thorough characterization of the nonlinear optical response of molecular gases and precise determination of $n_{2,el}$, γ , rotational and centrifugal distortion constants, dephasing rates, and rotational Raman spectrum in a self-consistent quantitative measurement.

2. Theory

The major mechanisms that cause NLR in dilute gases of linear molecules are the nearly instantaneous bound-electronic response, and the noninstantaneous induced rotational response that originates from the net alignment of the molecular ensemble. At equilibrium, the molecules are oriented randomly and the ensemble has zero net alignment, meaning the linear refractive index n_0 depends on the orientationally averaged polarizability. Changes in the degree of alignment lead to a refractive index change of [28, 29]

$$\Delta n_{rot}(t) = \frac{N\Delta\alpha}{2\epsilon_0 n_0} \left(\langle \cos^2 \theta(t) \rangle - \frac{1}{3} \right), \quad (1)$$

where N is the number density of molecules, $\Delta\alpha = \alpha_{\parallel} - \alpha_{\perp}$ is the difference in polarizabilities parallel and perpendicular to the molecular axis, ϵ_0 is the vacuum permittivity, n_0 is the real linear refractive index of the probe, θ is the angle between the molecular axis and the field polarization, and the angle brackets indicate ensemble averaging. A strong excitation pulse causes a time-dependent change in the refractive index experienced by a weak probe due to both mechanisms (see, for example, Refs [10, 30]. and [18] for a derivation) given by

$$\Delta n(t) = 2n_{2,el}I_e(t) + \frac{cN}{4}(\Delta\alpha)^2 \sum_J T_J \int_{-\infty}^t I_e(t') \sin(\omega_{J,J-2}(t-t')) e^{-\Gamma_{J,J-2}(t-t')} dt', \quad (2)$$

where $I_e(t)$ is the excitation (pump) irradiance, c is the speed of light in vacuum, $\omega_{J,J-2}$ and $\Gamma_{J,J-2}$ are the frequency difference and dephasing rate between rotational states J and $J-2$, respectively, and

$$T_J = \frac{2}{15\hbar} \frac{J(J-1)}{2J-1} (\rho_{J,J}^{(0)} - \rho_{J-2,J-2}^{(0)}) \quad (3)$$

is a weighting factor [10] that accounts for the initial occupation probability $\rho^{(0)}$ of the rotational states, which depends on both temperature and spin statistics of the atomic nuclei [31]. The relative occupation of even and odd rotational states is dictated by the spin of identical nuclei due to exchange symmetry, which is symmetric for integer spin, and antisymmetric for half integer spin [31].

The first term in Eq. (2) is the refractive index change due to the bound-electronic response, Δn_{el} . The factor of 2 in the first term arises from interference of the two beams doubling the phase change [32], sometimes referred to as weak-wave retardation [33]. This does not appear in the second term since the rotational response is too slow to follow the rapidly moving irradiance grating in nondegenerate experiments [23]. The bound-electronic contribution to the NLR originates from the transition dipole coupling of electronic states. It is characterized by the purely electronic second hyperpolarizability γ defined by [25]

$$\mu = \mu^{(0)} + \alpha E + \beta E^2 + \gamma E^3, \quad (4)$$

where μ is the dipole moment, expanded as a power series of the total electric field E , which is a sum of the pump and probe fields of the form $E_\theta(\omega)\cos(\mathbf{k}\cdot\mathbf{r}-\omega t)$, where $E_\theta(\omega)$ is the pulse envelope at central frequency ω , and \mathbf{k} is the wavevector. $\mu^{(0)}$ is the permanent dipole moment, α is the linear polarizability, and β is the first polarizability. In the symmetric molecules we investigate here, $\mu^{(0)}$ and β are zero. In lossless media, γ is related to the nonlinear refractive index by

$$n_{2,el} = \frac{3N}{4\epsilon_0^2 c n_0 n_{0,e}} f^{(3)} \text{Re}\langle\gamma\rangle, \quad (5)$$

where $n_{0,e}$ is the (real) linear refractive index of the excitation wavelength, $f^{(3)} = [(n_0^2 + 2)/3]^2 [(n_{0,e}^2 + 2)/3]^2$ is the Lorentz-Lorenz local field correction factor for third-order nonlinearities which is ~ 1 in dilute gases [28, 29], and angle brackets about γ indicate orientational averaging over the tensor elements [34].

The second term in Eq. (2) is the refractive index change due to the induced rotation of linear molecules, Δn_{rot} , where (classically) the excitation field induces a dipole along the molecular axis, applies a torque, and causes the molecules to rotate. The density matrix approach is applied to calculate the degree of alignment, where the quantization of rotational energies is considered [10, 18, 30]. States of total angular quantum number J have rotational energy $E_{rot} = hcBJ(J+1) - hcDJ^2(J+1)^2$, where h is Planck's constant, and B and D are the rotational and centrifugal distortion constants, respectively [31]. Using ultrashort femtosecond pulses, with a bandwidth similar to the initial rotational energy distribution, many rotational eigenstates are coherently excited via two-photon stimulated Raman transitions, and the resulting rotational superpositions have well-established phase relationships which evolve coherently at their Raman frequencies $\omega_{J,J-2}$. Quantum interference of the superposed rotational states leads to temporal recurrences in the degree of alignment of the molecular ensemble, which directly cause refractive index revivals (the second term in Eq. (2)). Because the frequencies $\omega_{J,J-2}$ are almost equally spaced, the constant phase

relationship imposed by the excitation results in periodic revivals. This is analogous to a mode locked laser, where the constant spectral phase across many longitudinal modes of a laser cavity results in a periodic pulsed output.

3. Beam deflection technique

The BD experimental setup is very similar to that reported previously [22, 23]. The excitation beam is focused to induce a change in the refractive index of the sample that follows its spatially Gaussian irradiance profile. The probe beam, of different wavelength than the excitation, is focused to a spot size $\sim 3\text{-}5 \times$ smaller than that of the excitation and is displaced from the excitation's center to where its irradiance gradient is maximized, as shown in Fig. 1(a). The probe beam then experiences the induced refractive index gradient, much like a prism, and is deflected by a small angle.

In gas phase measurements, the sample is thick and the interaction length is not as well characterized as it is with a thin solid sample. For the extended interaction length of the two beams in a gas, a three dimensional geometry for the beam overlap is implemented to maximize the deflection. The beams cross one another in the vertical yz -plane, as illustrated in Fig. 1(b), but are displaced in the horizontal xz -plane causing deflection in the x -direction, see Fig. 1(c). The crossing angle is kept small ($< 1^\circ$) to maximize the interaction length while still spatially separating the excitation from the probe at the detector. Deflection of the probe is detected by a quad-segmented Si photodiode (OSI QD50-0-SD) placed in the far-field. The difference in energy falling on the left (L) and right (R) sides of the segmented photodiode, $\Delta E = E_{left} - E_{right}$, is measured versus delay τ between the excitation and probe pulses using lock-in detection, which is normalized to the total energy E .

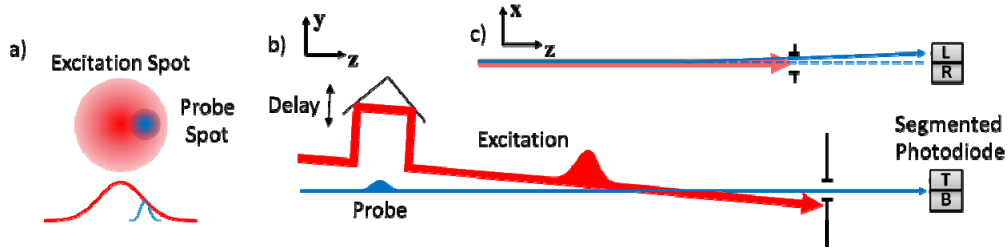


Fig. 1. (a) Excitation and probe irradiance distributions at the sample plane where the probe is displaced horizontally to the maximum irradiance gradient. (b) The excitation and probe crossing geometry in the yz -plane, and (c) displacement in the xz -plane resulting in a deflection in the x -direction at the segmented photodiode.

The rotational refractive index change occurs just after zero delay as well as at quarter multiples of the revival period $T = (2Bc)^{-1}$. Fitting is conducted by numerically evaluating Eqs. (2) and (3), considering both bound-electronic and molecular rotational contributions. The BD signal is then calculated as a function of delay, which for small signals is given by [22]

$$\frac{\Delta E(\tau)}{E} = \frac{2\sqrt{2}}{\sqrt{\epsilon\pi}} \frac{w_{0,p}}{w_e} k_{0,p} L \langle \Delta n(\tau) \rangle, \quad (6)$$

where $k_{0,p} = 2\pi/\lambda_p$, λ_p is the probe wavelength, L is the interaction length, $w_{0,p}$ and w_e are the probe and excitation spot sizes (HW1/e²M), and

$$\langle \Delta n(\tau) \rangle = \int_{-\infty}^{\infty} \Delta n(t) I_p(t - \tau) dt \Big/ \int_{-\infty}^{\infty} I_p(t) dt, \quad (7)$$

is the refractive index change temporally averaged over the probe irradiance $I_p(t)$. Equation (6), which is the same as Eq. (6) from [22], assumes a small interaction length such that the probe and excitation do not spatially walk off one another, and the phase change induced in the probe is uniform across the sample thickness. For our gaseous samples this assumption is no longer valid given the large thicknesses. To quantitatively measure the bound-electronic NLR of gases, we therefore perform relative measurements using the well-known values of $\Delta\alpha$, which have been measured from gas phase Raman spectroscopy [35, 36], as a self-reference. The relative contributions from the second hyperpolarizability γ and $\Delta\alpha$ can be obtained by evaluating the magnitudes of the bound-electronic and molecular rotational contributions in the measurement. It can be shown from Eqs. (2) and (5) that

$$\frac{\Delta n_{el}}{\Delta n_{rot}} \propto \frac{\gamma}{(\Delta\alpha)^2}. \quad (8)$$

Within one self-consistent BD measurement, where Δn_{el} and Δn_{rot} are separated, γ can be precisely determined by using known values of $\Delta\alpha$ as a reference. In this case, experimental uncertainties in the interaction length, number density, and excitation irradiance that affect absolute measurements are eliminated.

4. Air measurements

We first perform experiments on N_2 and O_2 in ambient air. A commercial Ti:sapphire laser system (Coherent Legend Elite Duo HE +) operating at 800 nm with 12 mJ output energy at a repetition rate of 1 kHz, is used for the excitation. A 30 cm motorized translation stage provides up to 2 ns of delay of the excitation pulse with respect to the probe pulse. The group velocity in air is accounted for in converting from stage position to temporal delay using $n_{0,e} = 1.000275$ [37]. Approximately 0.8 μJ of the laser output is focused into a 1 cm cuvette filled with water to generate a white-light continuum (WLC), from which the probe is spectrally filtered at 650 nm with a 40 nm bandwidth (FWHM). A polarizer is used to ensure linear polarization of the probe, and a half-wave plate controls the angle of its polarization with respect to that of the excitation. The excitation and probe beams are focused to spot sizes $w_e = 177 \mu\text{m}$ and $w_{0,p} = 60 \mu\text{m}$, and have pulse widths $\tau_e = 83 \text{ fs}$ and $\tau_p = 265 \text{ fs}$ (FWHM), respectively, as determined by auto- and cross-correlation. The peak excitation irradiance is 10 TW/cm^2 , which is below the filamentation threshold [38, 39].

Figure 2(a) shows the measured BD signal (black) of ambient lab air for a probe polarized parallel to the excitation over the first 28 ps of delay. A numerical fit (violet) using Eqs. (2) and (6) includes both N_2 and O_2 weighted by their relative atmospheric concentrations at 1 atm [40]. Figures 2(b) and 2(c) show the individual fits for N_2 (red) and O_2 (blue), respectively. The amplitude of the fit was matched to the data by allowing the interaction length L to vary, since the interaction length was not characterized. The best agreement was found with $L = 6.5 \text{ cm}$. We therefore obtain a relative fit, which gives a good determination of the ratio $\Delta\alpha(O_2)/\Delta\alpha(N_2)$, that agrees well with the values in [35] (see Table 1). The full revival for O_2 can be seen at 11.5 ps and for N_2 at 8.5 ps. Due to the contributions of both molecules to the total measured signal, certain time delays display features from the sum of each constituent. For example, the circled feature at 6 ps in Fig. 2 shows the contribution from the $\frac{1}{2}$ revival of O_2 and $\frac{3}{4}$ revival of N_2 .

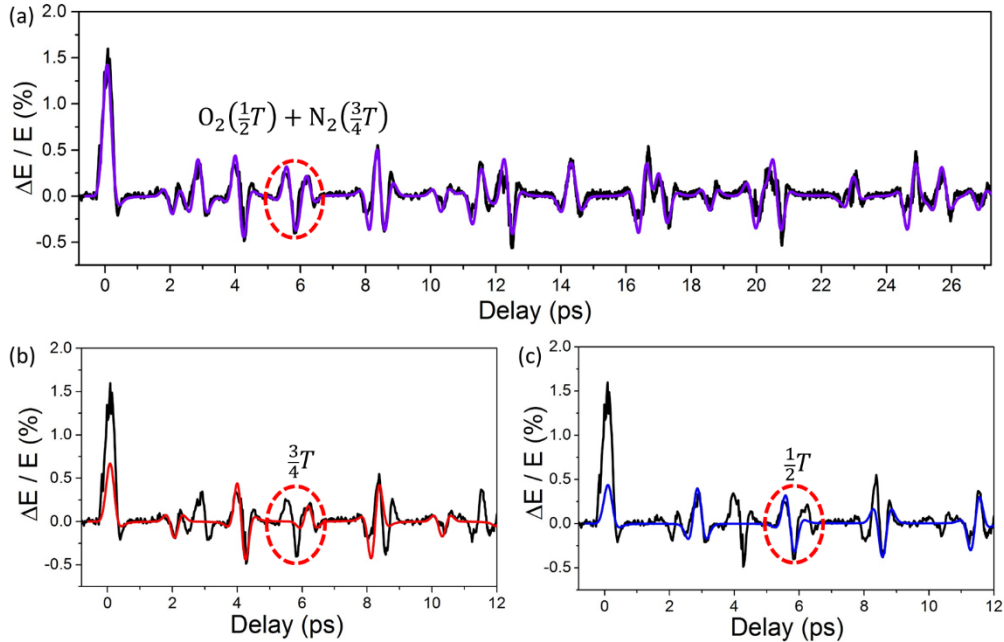


Fig. 2. (a) (black) Measured BD signal from air and (violet) fit, which is a sum of both the contribution of N_2 and O_2 . (b) and (c) show the first 12 ps of delay, which includes the first full revival periods, along with fits of N_2 (red) and O_2 (blue), respectively. The circled feature at about 6 ps highlights the combined effect of the $\frac{3}{4}$ revival of N_2 and $\frac{1}{2}$ revival of O_2 .

Nuclear spin statistics have an impact on the measured beam deflection signal. Both even and odd rotational states contribute equally to the $\frac{1}{2}$ and full revivals, but have the opposite sign for the $\frac{1}{4}$ and $\frac{3}{4}$ revivals [18]. The spin statistics of $^{16}O_2$ dictate that the rotational levels have only odd J values [31], and its $\frac{1}{4}$ and $\frac{1}{2}$ revivals therefore have equal amplitudes. $^{14}N_2$, however, has a 2:1 ratio of even and odd states [31], meaning the even states' contribution to the $\frac{1}{4}$ and $\frac{3}{4}$ revivals are partially canceled by the odd states'. The $\frac{1}{4}$ and $\frac{3}{4}$ revivals from N_2 therefore have half the amplitude of the $\frac{1}{2}$ and full revivals, and the opposite sign of those from O_2 .

To obtain information about the dephasing rate, the BD measurement was performed out to 300 ps of delay, shown in Fig. 3(a) (where the initial signal just after zero delay is not shown). Additionally, the rotational Raman spectrum is provided by the Fourier transform as shown in Fig. 3(b) along with the fits for N_2 (red) and O_2 (blue). On top of the thermal distribution, the consequences of nuclear spin statistics can be seen in the amplitude of the rotational Raman peaks. For example, the $^{14}N_2$ spectrum the intensity of adjacent peaks follows the 2:1 ratio of symmetric to anti-symmetric states [31].

The spectral lines for rotational transitions are evenly spaced until higher rotational levels where a centrifugal deformation lengthens the bond and decreases the rotational frequency. The resulting centrifugal distortion for higher rotational levels can be seen in Fig. 3(c). The fit for low rotational states with (solid lines) and without (dashed lines) D lie on top of each other. However, for higher rotational states there is significant discrepancy from the measurement when neglecting centrifugal distortion. The values of B and D found from fitting are given in Table 1 for each molecule, which shows a comparison to Raman scattering measurements [41].

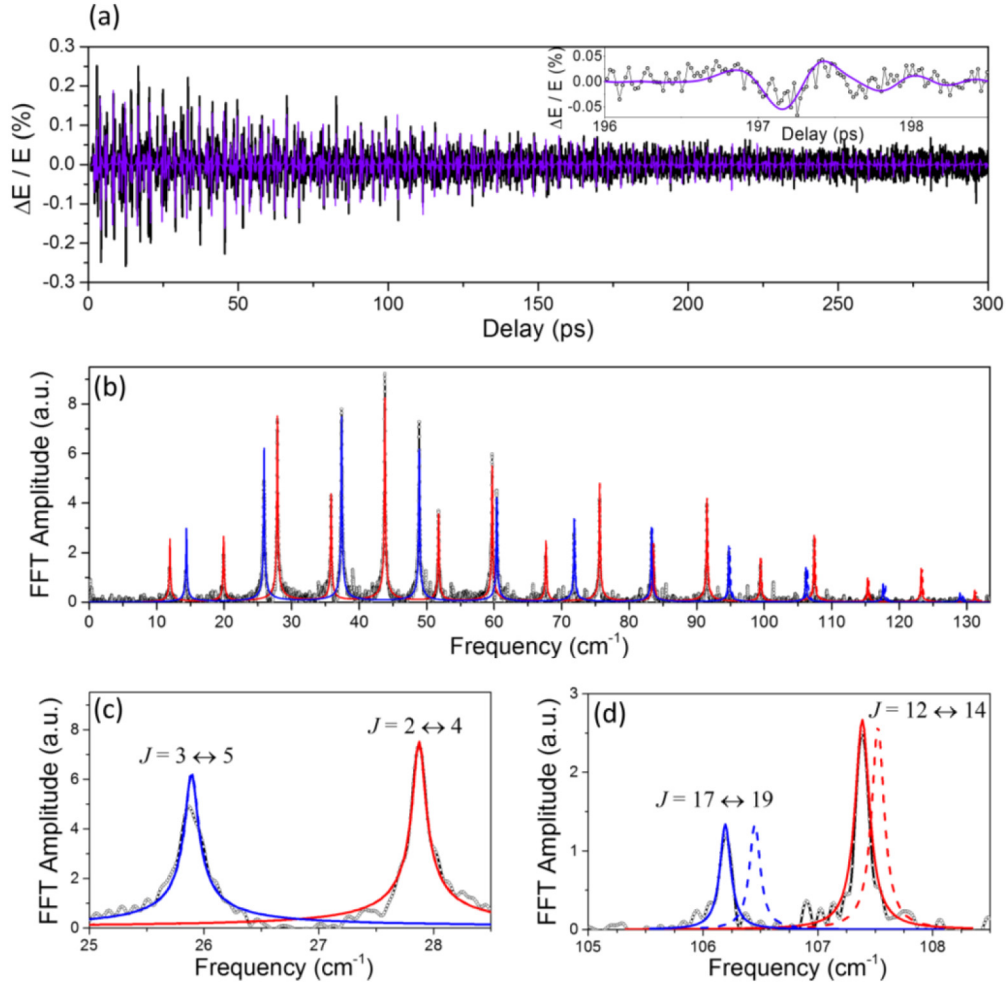


Fig. 3. (a) Measured (black) and fit (violet) BD signal of air out to 300 ps of delay where inset shows revival signal and fit about 197 ps. (b) Fourier transform of the BD signal for air (black) and fit for N_2 (red) and O_2 (blue). Zoomed in plots for (c) low and (d) high rotational states along with fits both neglecting (dashed) and including (solid) centrifugal distortion.

Table 1. Measured rotational and centrifugal distortion constants of N_2 and O_2

Molecule	$\Delta\alpha$ (10^{-40} Fm ²)	B (cm^{-1}) ^a	B (cm^{-1}) ^b	D (10^{-6} cm^{-1}) ^a	D (10^{-6} cm^{-1}) ^b
¹⁴ N_2	0.77 [35]	1.9896(2)	1.989574(12)	5.9(6)	5.76(3)
¹⁶ O_2	1.21 [35]	1.4377(2)	1.437682(9)	5.1(5)	4.852(12)

Closer examination of Fig. 3(b) reveals a J dependence of the linewidths, and therefore on the dephasing rate $\Gamma_{J,J-2}$. This is readily seen by comparing the widths of the low and high frequency peaks in Figs. 3(c) and 3(d), respectively. Figure 4 shows $\Gamma_{J,J-2}$ obtained from the rotational Raman spectrum by fitting each peak with a Lorentzian function versus J . The coherence of the rotational ensemble decays due to state-changing collisions, which requires energy exchange of $\Delta E_{J,J'} = |E_J - E_{J'}|$ (where $J \neq J'$) that increases with J [42–46]. Therefore higher rotational states are more resilient to collisions and have a narrower linewidth. Following [43], we fit the decay rates including only the minimum ΔJ (one for N_2 , two for

O₂) transitions in the exponential gap law, where the probability of collision transition is $P_{J, J'} = ae^{-cBJ}$, where a and c are fitting parameters.

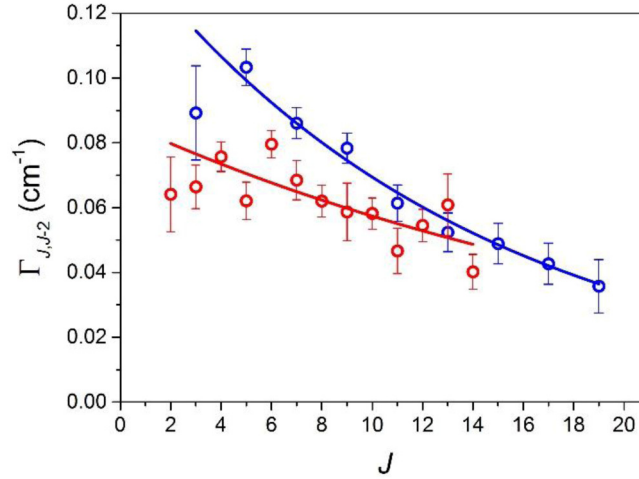


Fig. 4. Measurements (circles) of $\Gamma_{J,J'}$ versus J , for both N₂ (red) and O₂ (blue), and with exponential fits (curves).

The net alignment of the molecular ensemble creates a birefringence with an increased refractive index in the direction of alignment and a decreased refractive index in perpendicular directions. For some angle φ in between, the BD signal must go to zero. This is the “magic angle” ($\varphi = \arctan\sqrt{2} \approx 54.7^\circ$), where the rotational contribution to the BD signal is eliminated, leaving only the isotropic response which we attribute to the nearly instantaneous bound-electronic nonlinearity. We neglect the vibrational response since it is expected to be small [47], and our excitation pulse widths are too long (bandwidths too narrow) to excite vibrational modes, which we also do not observe in our experimental data. For the bound-electronic nonlinearity in the isotropic gas $\Delta n_{el,\perp} = \Delta n_{el,\parallel}/3$, and for rotational revivals $\Delta n_{rot,\perp} = -\Delta n_{rot,\parallel}/2$. We write the total index change as a sum of these two contributions, which for the three probe polarizations yields [22, 23]

$$\Delta n(0^\circ) = \Delta n_{el,\parallel} + \Delta n_{rot,\parallel}, \quad (9)$$

$$\Delta n(90^\circ) = \frac{1}{3} \Delta n_{el,\parallel} - \frac{1}{2} \Delta n_{rot,\parallel}, \quad (10)$$

$$\Delta n(54.7^\circ) = \frac{5}{9} \Delta n_{el,\parallel}. \quad (11)$$

Using a half-wave plate, the probe polarization angle with respect to that of the excitation was set to $\varphi = 0^\circ$ (parallel), 90° (perpendicular), and 54.7° (magic angle). Figure 5 shows the BD data for these polarization angles and fits using Eqs. (9)-(11). Because the amplitude of the rotational revival signal depends on the well-known values of $\Delta\alpha$, we use them as a self-reference and measure $n_{2,el}$ relative to $\Delta\alpha$. The fit in Fig. 5 corresponds to $n_{2,el}(\text{air}) = (10 \pm 2) \times 10^{-24} \text{ m}^2/\text{W}$ which agrees well with other measurements available from Refs [48, 49]. (see Table 2). Because there are multiple species in the gaseous mixture, we are unable to extract the second hyperpolarizability γ of the individual molecules. Instead, the average value for air is calculated with Eq. (5) to give $\gamma = (1.25 \pm 0.10) \times 10^{-62} \text{ C}^4\text{m}^4/\text{J}^3$, and compared to the concentration weighted average value $\gamma(\text{air}) = 0.78\gamma(\text{N}_2) + 0.21\gamma(\text{O}_2) + 0.01\gamma(\text{Ar})$ (using $N =$

$2.5 \times 10^{19} \text{ cm}^{-3}$, see Table 2). We find agreement, to within measurement error, with values determined from Shelton and Rice [26] when including dispersion by their Eqs. (18) and (19).

Table 2. Measured $n_{2,el}$ and weighted average value γ of air.

	This work	Wahlstrand <i>et al.</i> [48]	Liu and Chin [49]	Shelton and Rice [26]
$n_{2,el} (10^{-24} \text{ m}^2/\text{W})$	10 ± 2	7.9 ± 1.0	9.6	9.1
$\gamma (10^{-62} \text{ C}^4 \text{ m}^4/\text{J}^3)$	1.25 ± 0.20	0.98 ± 0.10	1.2	1.14
$\gamma (10^{-37} \text{ esu})$	10.1 ± 2.0	7.9 ± 1.0	9.7	9.2

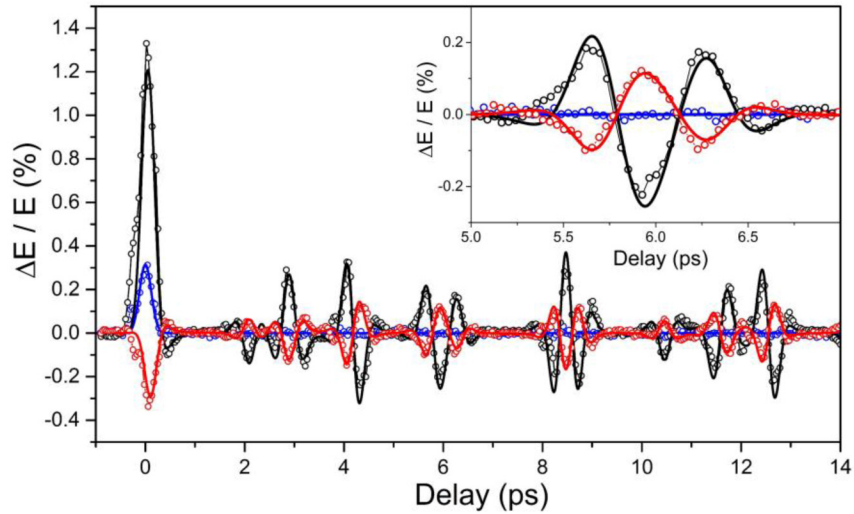


Fig. 5. (points) Measured and (curves) fit polarization dependence of NLR of air, including (black) parallel, (red) perpendicular, and (blue) magic angle polarizations. Inset shows zoomed in region about 6 ps of delay.

5. CS₂ measurements

Measurements of gaseous CS₂ are performed by filling a 10 cm borosilicate cell $\sim 1/3$ full with the liquid CS₂ (Sigma-Aldrich, 270660, $\geq 99.9\%$) and heating the cell windows to approximately 50 °C with a cell heater (Thorlabs GCH25-75) to prevent condensation, such that the upper portion of the cell is filled with gaseous CS₂. The sample cell is positioned such that the beams cross above the liquid CS₂ surface. A 1250 nm wavelength excitation pulse is generated from an optical parametric amplifier (Light Conversion, HE-TOPAS) pumped by the same laser source, and a WLC is generated in a 5 mm sapphire plate and spectrally filtered to give a 950 nm wavelength (FWHM 10 nm) probe. The beams are focused to $w_{0,p} = 90 \mu\text{m}$ and $w_e = 468 \mu\text{m}$, and have pulse widths $\tau_p = 137 \text{ fs}$ and $\tau_e = 120 \text{ fs}$. The excitation energy is 62 μJ , giving a peak irradiance of 140 GW/cm^2 . Equation (2) shows that Δn depends on $(\Delta\alpha)^2$, which for N₂ and O₂ is approximately two orders of magnitude less than that for CS₂ [36]. Thus, we do not observe any contribution from air in the measurements of gaseous CS₂, and an irradiance nearly two orders of magnitude greater was used for the air measurements.

Figure 6 shows BD measurements for CS₂. The larger moment of inertia of CS₂ compared to both N₂ and O₂ gives it a much longer revival period (152.4 ps). Only up to the first $3/4$ revival is shown, since the first full revival is below the noise floor, owing to the more rapid decay rate. Figure 6(c) shows the $1/2$ revival of CS₂ about 76 ps, plus an additional feature at a slightly longer delay (~ 79 ps), which is enlarged in Fig. 6(d). This feature is due to the sulfur isotopologue C³²S³⁴S, which has a greater moment of inertia than C³²S₂, and is the second

most common sulfur isotopologue. Based on this feature we measure a relative abundance of $(9 \pm 3) \%$, which agrees with the natural abundance of 8.6% [43]. This additional feature only appears for the $\frac{1}{2}$ revival, but is not observed at the $\frac{1}{4}$ or $\frac{3}{4}$ revivals. Because its two S isotopes are distinguishable, $C^{32}S^{34}S$ lacks nuclear exchange symmetry, meaning even and odd rotational states J are equally weighted [43]. Even and odd J states contribute equal and opposite signals at the $\frac{1}{4}T$ and $\frac{3}{4}T$ times, which cancel one another.

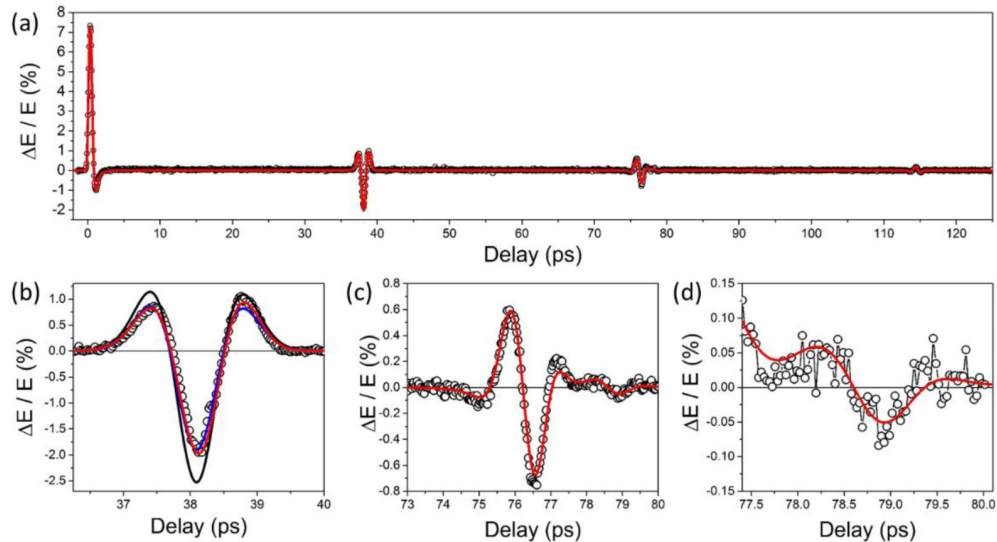


Fig. 6. (a) Measured (circles) BD signal from gaseous CS_2 and fit (red curve) including $C^{32}S^{34}S$ and the ground and first excited rotational states of $C^{32}S_2$. (b) $\frac{1}{4}$, and (c) $\frac{1}{2}$ revival $C^{32}S_2$. (d) shows the $\frac{1}{2}$ revival from $C^{32}S^{34}S$. In (b) both the black and blue curves neglect centrifugal distortion, and the black curve also assumes only ground state $C^{32}S_2$.

Additionally, at room temperature 15% of the $C^{32}S_2$ molecules are in the first excited vibrational state, the doubly-degenerate bending mode at $\nu_2 = 397 \text{ cm}^{-1}$ [43]. The two modes oscillate in the plane either perpendicular or parallel to the axis of rotation, which are symmetric and antisymmetric upon nuclear interchange, respectively, and therefore have opposite spin statistics [50]. We treat the modes with a single rotational constant and equal weighting of even and odd J states. Therefore, only $C^{32}S_2$ in the ground vibrational state has unequal weighting of even and odd J states, and is the only species that contributes to the $\frac{1}{4}$ and $\frac{3}{4}$ revivals. This leads to a greater relative amplitude of the $\frac{1}{2}$ revivals, since all molecules contribute. The impact of this can be seen in calculations of the $\frac{1}{4}$ revival shown in the black curve Fig. 6(b), which assumes all of the molecules are $C^{32}S_2$ in its ground vibrational state. The fit matches the amplitude of the signals at zero delay and at the $\frac{1}{2}$ revival (same as red curve in Fig. 6(c)) but clearly overestimates the $\frac{1}{4}$ revival signal. The rotational constant of the first excited vibrational state was found by giving the best fit with the $\frac{1}{2}$ revival. Centrifugal distortion is also included causing the asymmetries in the revival signals which can be seen, for example, in the $\frac{1}{4}$ revival where the peak at 38.8 ps is larger than the peak at 37.4 ps. For comparison, the blue curve in Fig. 6(b) neglects centrifugal distortion. With only knowledge of the energy and spin statistics of vibrational modes, the rotational and centrifugal distortion constants (given in Table 3) and relative abundance of isotopologues are found from the fit, which agree with Refs [14, 43, 51]. For comparison, Table 3 also shows values of B and D from Ref [43].

Table 3. Measured rotational and centrifugal distortion constants of CS₂

Molecule	$\Delta\alpha$ (10^{-40} Fm ²)	B (cm ⁻¹) ^a	B (cm ⁻¹) ^b	D (10^{-6} cm ⁻¹) ^a	D (10^{-6} cm ⁻¹) ^b
C ³² S ₂	10.53 [36]	0.10942(10)	0.1091271(6)	0.010(5)	0.01173(1)
C ³² S ₂ (ν_2)	10.53 [36]	0.10957(10)	0.10930(1) ^c	0.010(5)	0.01111(17) ^c
			0.10938(1) ^d		0.01157(13) ^d
C ³² S ³⁴ S	10.53 [36]	0.10617(10)	0.1059086(7)	0.010(5)	0.01105(12)

Figure 7 shows both measurements and fits of the signal from CS₂ about zero delay for (black) parallel, (red) perpendicular, and (blue) magic angle polarizations. Since we now have only a single molecular species contributing to the BD signal, we can extract the orientationally averaged γ . Using the known value of $\Delta\alpha$ given in [36], we fit the measurement at the magic angle for the bound-electronic nonlinearity and use Eq. (5) to find $\gamma(\text{CS}_2 \text{ gas}) = (19 \pm 4) \times 10^{-62} \text{ C}^4\text{m}^4/\text{J}^3$ ($(15 \pm 4) \times 10^{-37}$ esu). This capability to determine γ is of interest for theoretical purposes to compare calculated values of γ , which are typically performed for isolated molecules [25, 26]. Additionally, if we assume Lorentz-Lorenz local field correction factor [16, 17] we find agreement between this measurement and that in liquid CS₂ from [23] at the same average photon energy (1.15 eV, 1080 nm), where $\gamma(\text{CS}_2 \text{ liquid}) = (23 \pm 5) \times 10^{-62} \text{ C}^4\text{m}^4/\text{J}^3$ ($(18 \pm 4) \times 10^{-37}$ esu). Table 4 shows a comparison of these results, along with ratios of $\gamma(\text{CS}_2)$ to both $(\Delta\alpha)^2$ and $\gamma(\text{N}_2)$ [26]. In liquid phase $f^{(3)} = 5.35$ [52, 53], meaning local field effects have a significant impact on $n_{2,el}$, (see Eq. (5)).

Table 4. Comparison of γ of CS₂ from gas and liquid measurements.

	Gas Phase	Liquid Phase [23]
$\gamma / (\Delta\alpha)^2$ (10^{16} J^{-1})	17 ± 4	--
γ ($10^{-62} \text{ C}^4\text{m}^4/\text{J}^3$)	19 ± 4	23 ± 5
γ (10^{-37} esu)	15 ± 3	18 ± 4
$\gamma(\text{CS}_2) / \gamma(\text{N}_2)$	17 ± 4	20 ± 4

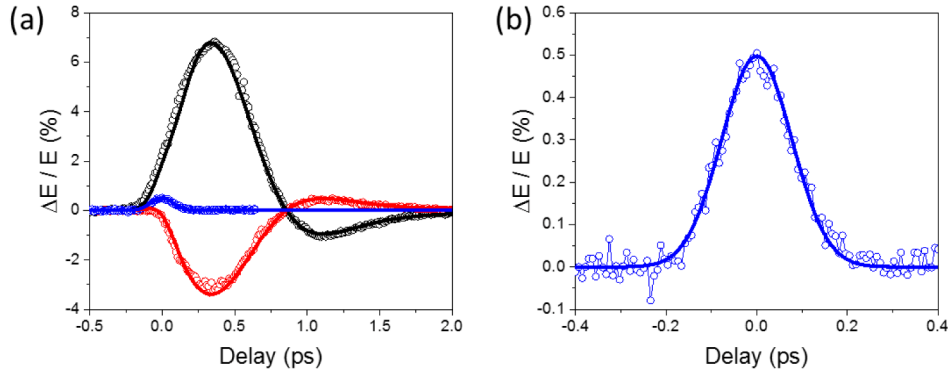


Fig. 7. (a) BD (circles) measurements of CS₂ about zero delay with (curves) fits for (black) parallel, (red) perpendicular, and (blue) magic angle polarizations. (b) Zoomed in measurement at the magic angle.

6. Conclusion

We demonstrate our recently developed BD technique's ability to measure NLR in dilute gases of air and CS₂ due to both the bound-electronic and coherent rotational responses. We identify the contributions from individual species that give rise to the total signal of a gaseous

mixture by separating the contributions to the air signal from N₂ and O₂. Since BD allows measurements out to long delays, collisional dephasing back to thermal equilibrium can be seen, and the Fourier transform of the measured signal yields the rotational Raman spectrum. From measurements of gaseous CS₂, we identify the signal from two isotopologues, C³²S₂ and C³²S³⁴S, and the first excited vibrational state. Based on these measurements, we are able to determine the rotational and centrifugal distortion coefficients, J dependent dephasing rates, and relative isotopologue concentration.

The bound-electronic response is separated from rotational revivals by taking advantage of their different polarization dependencies using the magic angle. With well-known values of $\Delta\alpha$, the signal from the coherent rotational revivals provides a self-reference, from which $n_{2,el}$ of air and orientationally averaged γ of CS₂ are unambiguously determined. Determination of γ of N₂ and O₂ is also possible by measuring pure gases. Measurement of γ of isolated molecules is important for comparison to theoretical models which are typically compared to measurements performed on liquids [34]. Furthermore, we have experimentally verified that Lorentz-Lorenz local field correction yields agreement between gas and liquid phase measurements of γ .

The BD technique may be applied to study the structure and dynamics of a variety of molecular species with anisotropic polarizabilities in gaseous mixtures. It may be further improved upon to provide absolutely calibrated measurements by fully characterizing the beam interaction geometry, or by incorporating a gas jet within a vacuum chamber such that the induced deflection is uniform over a well-defined interaction length. The latter has been implemented by Wahlstrand *et al.* in their single-shot spectral interferometry technique [39, 48, 54, 55], where measurement of a thin (~2 mm) laminar gas flow within a chamber at low background pressure allows absolutely calibrated measurements. Experimental investigations of other gases may help validate and improve theoretical models of γ and the limitations of the Lorentz-Lorenz local field correction.

Acknowledgments

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