



HfO₂/SiO₂ anti-reflection films for UV lasers via plasma-enhanced atomic layer deposition



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ABSTRACT

The laser-induced damage threshold (LIDT) of HfO₂/SiO₂ anti-reflection (AR) films for ultraviolet (UV) lasers was improved via low-temperature plasma-enhanced atomic layer deposition (PEALD). Focused on the chemical composition, optical absorption, surface scattering, and laser-resistance, the impact of precursor exposure time on PEALD SiO₂ film properties and growth temperature on PEALD SiO₂ and HfO₂ film properties were investigated respectively. When irradiated by UV laser, PEALD SiO₂ film exhibits a higher LIDT than the PEALD HfO₂ film, which is consistent with their less impurity content and lower absorption. A bilayer structure HfO₂/SiO₂ AR film for 355 nm laser was designed and experimentally demonstrated via PEALD growth at a temperature of 150 °C. The prepared PEALD AR film shows a reflectance <0.2% at 355 nm as designed and better laser-damage resistance with a LIDT of 24.4 J/cm² (355 nm, 7.8 ns) than the conventional e-beam deposition method.

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1. Introduction

High-power lasers have attracted growing interests in recent years for their applications in many areas, including clean energy [1], light-matter interaction [2], and communications [3]. Developing new thin-film optics that can endure in enormously high-power lasers is one of the main challenges [4].

Traditional high-resistance laser coatings were mostly prepared by electron-beam evaporation [5,6], ion-assisted deposition [7,8], ion-beam sputtering [9], and sol-gel method [10]. In recent years, atomic layer deposition (ALD) shows exceptionally precise

thickness controllability [11], high uniformity [12], and unparalleled conformality [13]. Many works [14,15] have studied the influence of thermal ALD deposition parameters on the growth behavior, chemical composition, and electrical property. ALD also shows a high laser-induced damage threshold (LIDT) [16] and is a promising method for preparing thin films for high-power lasers applications [16,17]. Current attempts to prepare thin films for high power lasers via ALD focus on the use of TiO₂/Al₂O₃ or HfO₂/Al₂O₃ layers for near-infrared (NIR) anti-reflection (AR) films [17–20]. A high quality 1064 nm HfO₂/Al₂O₃ AR film prepared by ALD with an LIDT value of 30 J/cm² was reported by Zhang et al. [17]. There are relatively few reports on the use of ALD technology to prepare UV AR films. Most optical materials have large absorption in the UV region due to the limitation of optical bandgap width [21] and are sensitive to the absorption precursors such as defects [22] and impurities [23]. Also, nodules inside films have higher electric field intensification at the UV region [24], which is harmful to laser damage resistance of films [25]. Therefore, it is more difficult to obtain a high LIDT AR film in the UV region compared to the NIR

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region. Recently, a 355 nm AR film grown from HfO₂/Al₂O₃ layers with an LIDT of 4.5 J/cm² was reported by Hao Liu et al. [20]. However, the TiO₂/Al₂O₃ or HfO₂/Al₂O₃ design structures are not the best strategies for AR films for lasers in the UV to NIR wavelength region. According to a worldwide laser damage competition for UV AR films, HfO₂/SiO₂ films show the highest LIDT other than sol-gel SiO₂ films [26]. However, the sol-gel film suffers from chemical instability in ambient conditions [27,28]. SiO₂ is an ideal low refractive index material not only for coating quantum dots [29,30], but also for high-power laser applications from UV to NIR wavelength region because of its low loss and large optical bandgap properties [31,32]. Unfortunately, temperatures higher than 250 °C required for thermal ALD growth of SiO₂ [33,34] cause polycrystallization of HfO₂ film [35] and the shifting of sub-surface impurities to the surface of the substrate [36], which are not conducive to obtain AR film with high LIDT. Plasma-Enhanced ALD (PEALD), which provides similar growth characteristics as thermal ALD [37], has the advantage of low-temperature growth. It can increase the growth rate [12], better control the stoichiometry of materials [34,38], and allow larger freedom in processing conditions for tuning the properties of thin films [39]. We previously investigated the non-uniformity, surface roughness and simply compared the LIDT of PEALD SiO₂ and HfO₂ films [40], but further research is needed to study the performance of SiO₂ and HfO₂ films concerning their applications in laser systems. In this work, we aim to improve the LIDT of UV AR films via low-temperature PEALD growth of HfO₂/SiO₂ layers. In this method, O₂/Ar plasma is used to obtain high plasma density and achieve low-temperature growth and high growth rate [12]. The influence of precursor exposure time on growth characteristics of PEALD SiO₂ films was studied. The effects of growth temperature on the properties of both PEALD SiO₂ and HfO₂ films were investigated, especially the chemical composition, optical absorption, and laser damage resistant properties. A 355 nm AR film was then prepared for UV laser application via PEALD, the LIDT and damage mechanism were tested and analyzed.

2. Experimental

2.1. Preparation of PEALD SiO₂ and HfO₂ films

A commercial PEALD apparatus with an integrated remote plasma source (Picosun Advanced R200, Finland) was utilized. SiO₂ and HfO₂ films were deposited on a fused silica substrate or a BK7 glass substrate. The BK7 glass substrate was used to characterize the thicknesses of the film, while the fused silica substrate was used to investigate other properties of the film. To obtain a clear surface of the substrate, the fused silica substrate was first etched in a mixture solvent of ~1% HF and ~15% NH₄F for 8 min, then washed with deionized water in an ultrasonic bath for 5 min, and finally dried with an oven lamp. The BK7 glass substrate was cleaned using the last two steps of the cleaning procedure for a fused silica substrate.

Table 1
Deposition parameters for PEALD AR film.

| | | HfO ₂ | SiO ₂ |
|-------------------------|-----------------------|------------------|------------------|
| Exposure time (s) | Precursor | 1.6 | 0.4 |
| | Plasma O ₂ | 11 | 11 |
| Purge time (s) | Precursor | 19 | 19 |
| | Plasma O ₂ | 10 | 8 |
| Boosting time (s) | | 1.7 | / |
| Picoflow time (s) | | / | 11 |
| Growth temperature (°C) | | 150 | 150 |
| Number of cycles | | 230 | 698 |

The growth of SiO₂ and HfO₂ via PEALD was performed by alternately exposing the corresponding precursors - trisdimethylamino silane (SiH(N(CH₃)₂)₃, 3DMAS) and tetrakisethylmethylamino hafnium (Hf(N(CH₃)(CH₂CH₃)₄, TEMAH), and O₂/Ar gas mixture plasma reactant. 3DMAS and TEMAH were injected into the chamber through the carrier gas N₂. The temperature of the 3DMAS source was maintained at room temperature, while the temperature of the TEMAH source was maintained at 120 °C. Three groups of films were prepared for comparative studies. Groups G1 and G2 are SiO₂ films, consisting of 750 SiO₂ growth cycles grown at different temperatures (in the range of 50–200 °C) and different exposure times (in the range of 0.1–0.7 s), respectively. The sequence of pulses for one cycle deposition of the SiO₂ film was as follows: 3DMAS feeding (0.1 s or 0.1–0.7 s), N₂ purging (19 s), O₂ feeding (11 s) with RF plasma power of 2500W, and Ar purging (8 s). The exposure time of the 3DMAS precursor of the SiO₂ films in group G1 was 0.1 s, and the deposition temperature of the SiO₂ films in group G2 was 150 °C. Group G3 is HfO₂ films, consisting of 500 cycles grown at different temperatures (in the range of 50–200 °C). The sequence of pulses for one cycle deposition of the HfO₂ film was as follows: TEMAH feeding (1.6 s), N₂ purging (19 s), O₂ feeding (11 s) with 2500W RF plasma power, and Ar purging (10 s).

2.2. Preparation of PEALD and E-beam 355 nm AR films

355 nm AR films were prepared using PEALD and e-beam deposition for comparative studies. The AR film was grown on fused silica substrates that were cleaned using the cleaning procedure of the three steps described above. The recipe of the PEALD AR film is given in terms of PEALD cycle sequences and shown as follows: 230 cycles × HfO₂ + 698 cycles × SiO₂, thereby obtaining a HfO₂ layer with a designed thickness of 20.1 nm and a SiO₂ layer with a designed thickness of 76.1 nm. The exposure time of 3DMAS precursor of SiO₂ film was 0.4 s, the growth temperature was 150 °C. Other growth parameters were consistent with the growth parameters of SiO₂ and HfO₂ films. Table 1 shows further deposition details.

The coating structure and layer thickness of the e-beam deposited AR film are the same as those of PEALD AR film. The coating chamber was heated to 150 °C before deposition. HfO₂ and SiO₂ layers were evaporated from metal Hf and SiO₂ materials at the deposition rates of 0.04 nm/s and 0.6 nm/s, respectively. The oxygen pressure of HfO₂ and SiO₂ were 2.8 × 10⁻² Pa and 5 × 10⁻³ Pa, respectively.

2.3. Thin-film characterizations

The thickness and refractive index of the SiO₂ and HfO₂ films were determined by using a spectroscopic ellipsometer (Horiba Uvisel 2) and the Tauc-Lorentz model of the DeltaPsi2 software. The scattering of SiO₂ and HfO₂ films at 632.8 nm was measured by a

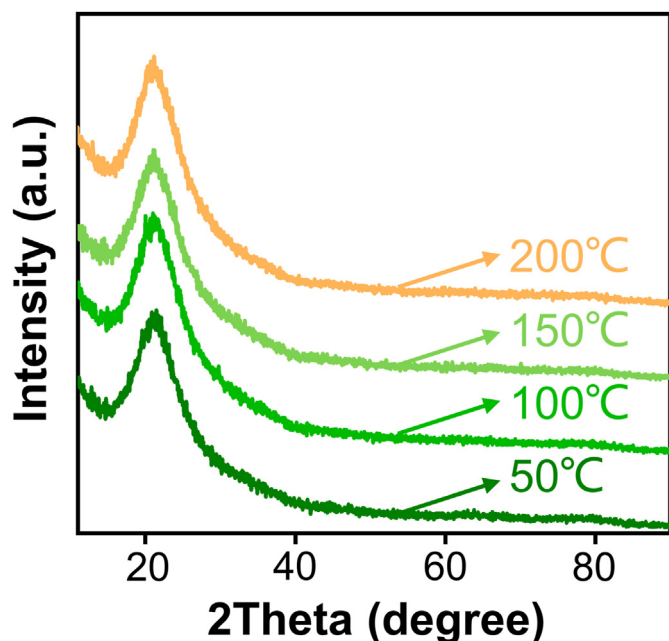


Fig. 1. XRD spectra of PEALD SiO_2 films grown at different temperatures.

home-made system based on a total integrated scattering method [41]. The structural information was measured by X-ray diffraction (XRD, Panalytical Empyrean). The film absorption at 355 nm was measured by a home-made system based on the surface thermal lensing technique [42]. The 355 nm laser power irradiated on the film was in the range of 99.5–291.7 W/mm^2 . The absorption value of each SiO_2 film was recorded for 600 s, and the average value was taken as the absorption. For HfO_2 and 355 nm AR films, the

absorption value decreases with the laser irradiation time before reaching a stable value. The absorption value of each HfO_2 film and 355 nm AR film was recorded for 2000 s, and the average value at the stable stage was taken as the absorption. The chemical composition of the SiO_2 and HfO_2 films was analyzed by an X-ray photoelectron spectroscopy (XPS, Thermo Scientific) equipped with a monochromatic $\text{Al K}\alpha$ (1486.6 eV) X-ray source. To remove the surface layer which adsorbs water and oxygen in the air, the coating was etched for 80 s with 1 keV Ar^+ ions before the XPS scan. The binding energy of the C 1s line is taken as 284.8 eV for calibrating the obtained spectra. The reflectance spectrum was measured by a UV–visible spectrometer (PerkinElmer Lambda 1050) at normal incidence. The 1-on-1 LIDT was measured according to ISO 21254 using a Gaussian-shape 3ω Nd:YAG laser (355 nm, 7.8 ns). The LIDT test was performed under normal incidence, and the effective spot area on the sample surface was 0.28 mm^2 . The maximum fluence with zero probability of laser damage was determined as LIDT. The surface and cross-section morphologies of the laser-induced damaged site were obtained by a focused ion-beam scanning electron microscope (FIB-SEM, Carl Zeiss AURIGA CrossBeam). The chemical composition of the laser-induced damaged site was analyzed by an energy dispersive spectroscopy (EDS, Oxford X-Max, 50 mm^2), operated at an accelerating voltage of 15 kV. Before FIB-SEM and EDS characterization, a 20 nm-thick Cr film was deposited on the sample surface by ion beam sputtering (Quorum Q150T ES).

3. Results and discussions

3.1. Properties of the SiO_2 films in group G1

The ellipsometrically determined thicknesses of SiO_2 films grown at 50, 100, 150, and 200 $^\circ\text{C}$ are 88.1, 83.9, 83.2, and 73.7 nm, respectively. Fig. 1 shows XRD spectra of the SiO_2 films, and no

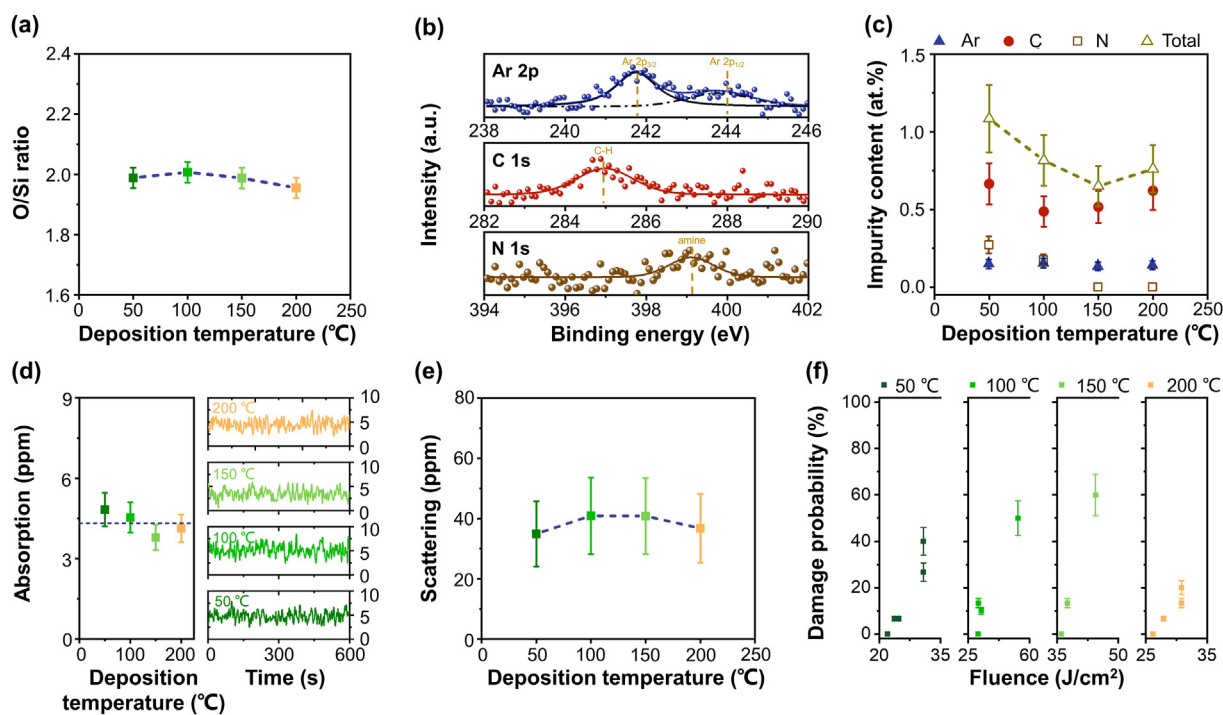


Fig. 2. (a) O/Si ratio of PEALD SiO_2 films in Group G1. (b) A typical XPS spectrum of Ar 2p, C 1s, and N 1s in a PEALD SiO_2 film grown at 100 $^\circ\text{C}$. (c) Individual and total impurity contents of Ar, C, and N elements, (d) absorption measured at 355 nm, (e) scattering measured at 632.8 nm, and (f) laser-induced damage probability for PEALD SiO_2 films in Group G1. The dash-dot line in (d) shows the average absorption of PEALD SiO_2 films grown at different temperatures.

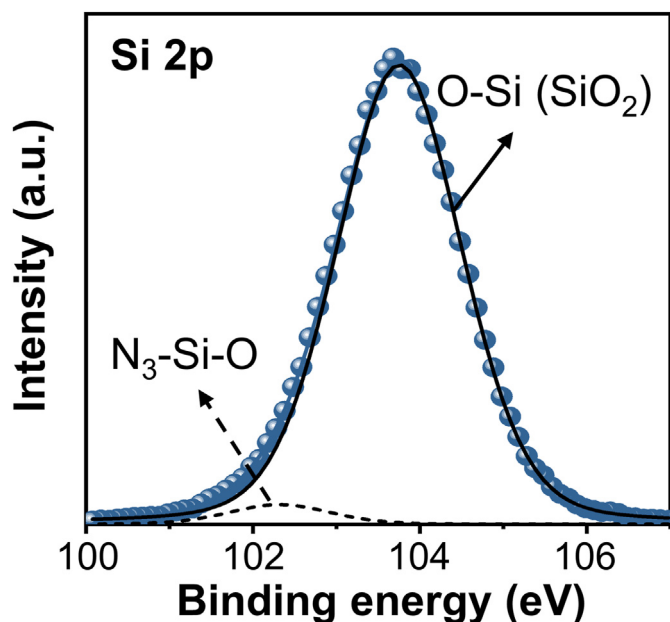


Fig. 3. A typical Si 2p spectrum of a PEALD SiO₂ film deposited at a temperature of 100 °C.

peak is observed except for a broad peak corresponding to the fused silica substrate, which indicates that the SiO₂ film is amorphous. XPS was used to examine the chemical composition and bonding structure of the SiO₂ films. The O/Si ratio determined by XPS is shown in Fig. 2a, and the O/Si ratio of the SiO₂ film

deposited at 50, 100, and 150 °C is close to the stoichiometric ratio of 2. As the deposition temperature further increases, the O/Si ratio of the PEALD SiO₂ film decreases. Ar and C elements are observed in all SiO₂ films, but N element is only observed in SiO₂ films deposited at temperatures of 50 and 100 °C. Fig. 2b shows typical XPS spectra of Ar, C, and N in a SiO₂ film grown at a temperature of 100 °C. Two noticeable peaks, at 241.8 eV (Ar 2p_{3/2}) and 243.9 eV (Ar 2p_{1/2}), are observed in the Ar 2p spectrum, which is in agreement with previously reported binding energies [43]. The peak at 284.9 eV in the C 1s spectrum is related to the C–H bond [44]. The peak at 399.1 eV in the N 1s spectrum is associated with the undoped amine unit [45], which comes from the 3DMAS precursor. A typical Si 2p spectrum of a SiO₂ film deposited at a temperature of 100 °C is shown in Fig. 3 and can be fitted with two Gaussian peaks. The binding energy of Si 2p peak located at 103.7 eV corresponds to the Si–O bond of SiO₂ [46]. The lower binding energy of Si 2p peak located at 102.3 eV corresponds to the N₃–Si–O bond [47], indicating the presence of residual 3DMAS in the SiO₂ film. The individual and total contents of the Ar, C, and N elements in the SiO₂ films are shown in Fig. 2c. As the deposition temperature increases, the total impurity content decreases first and then increases. The SiO₂ film deposited at a temperature of 150 °C shows the lowest impurity content. The absorption and scattering of SiO₂ films are shown in Fig. 2d and e, respectively. All SiO₂ films show pretty low absorption, which fluctuates with the measurement time. The scattering of SiO₂ film is relatively low and corresponds to the low roughness of SiO₂ film [40]. The laser-induced damage probability of the SiO₂ films is compared in Fig. 2f. The SiO₂ film deposited at a temperature of 150 °C exhibits the highest LIDT, which is consistent with the lowest impurity content observed.

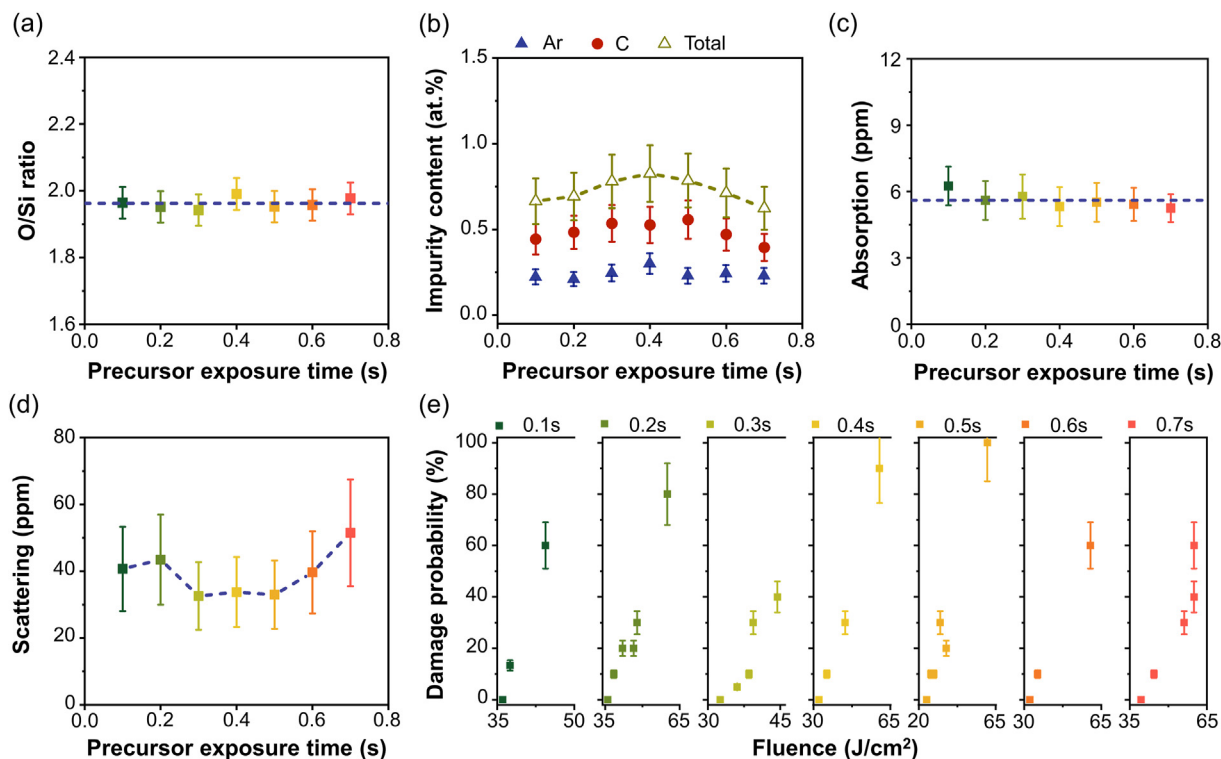


Fig. 4. (a) O/Si ratio, (b) individual and total impurity contents of Ar and C, (c) absorption measured at 355 nm, (d) scattering measured at 632.8 nm, and (e) laser-induced damage probability for PEALD SiO₂ films in Group G2. The dash-dot lines in (a) and (c) show the average O/Si ratio and absorption of PEALD SiO₂ films grown at different precursor exposure times.

3.2. Properties of the SiO₂ films in group G2

The ellipsometrically determined thicknesses of SiO₂ films grown at 0.1, 0.2, 0.3, 0.4, 0.5, 0.6 and 0.7 s are 69.0, 73.9, 76.5, 77.1, 77.8, 77.6 and 75.1 nm, respectively. The XPS determined O/Si ratio of the SiO₂ films is close to the stoichiometric ratio of 2, as shown in Fig. 4a. Ar and C elements are observed in all SiO₂ films prepared with different precursor exposure times, while the N element is not observed. The individual and total impurity contents of Ar and C of the SiO₂ films are shown in Fig. 4b. When the precursor exposure time is longer than 0.4 s, the impurity content decreases slightly with the increase of the precursor exposure time. The absorption and absorption curve versus the measurement time of the SiO₂ films are shown in Figs. 4c and 5. The absorption of SiO₂ films is relatively low and fluctuates around 5.6 ppm with increasing precursor exposure time. Compared with SiO₂ films deposited with other precursor exposure times, the SiO₂ film deposited with the precursor exposure time in the range of 0.3–0.5 s shows lower scattering, as shown in Fig. 4d. Light scattering normally increases with surface roughness [48]. The lower scattering of the SiO₂ film deposited with the precursor exposure time in the range of 0.3–0.5 s corresponds to the lower surface roughness in our previous work [40]. Fig. 4e shows the laser-induced damage

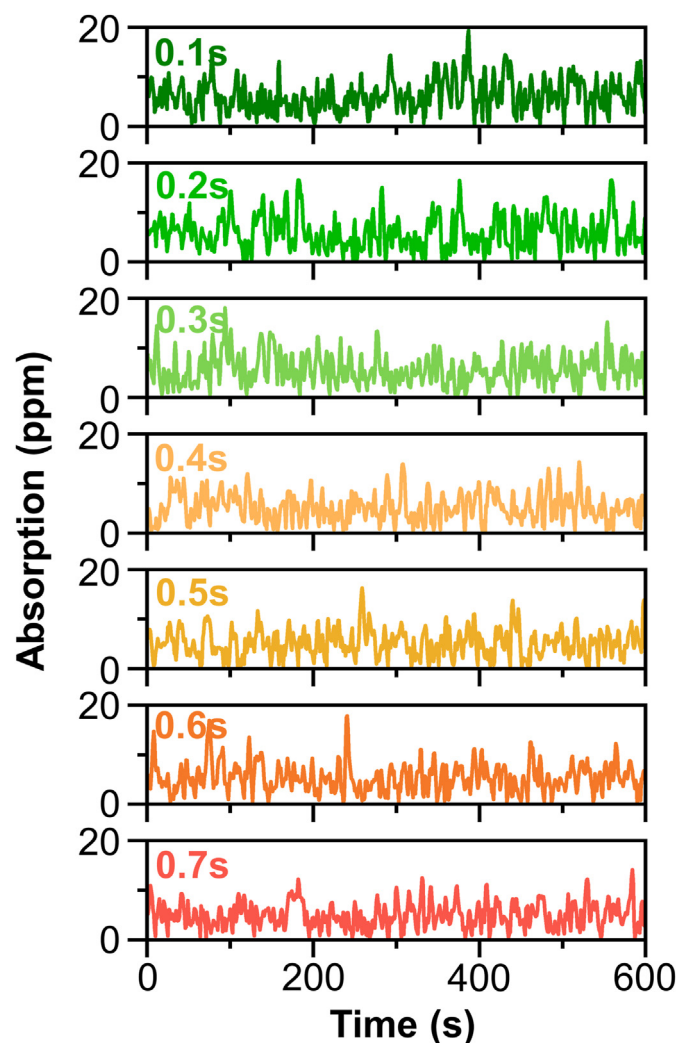


Fig. 5. The absorption curves at 355 nm versus measurement time of PEALD SiO₂ films grown at different precursor exposure times.

probability versus the input fluence of the films deposited at different precursor exposure times, indicating that the PEALD SiO₂ films grown at precursor exposure times in the range of 0.4–0.6 s exhibit relatively lower LIDT, which is due to the higher impurity content and therefore more absorption defects [23]. Although the LIDT of a PEALD SiO₂ film grown with a precursor exposure time of 0.4 s is slightly lower than the highest observed LIDT, the precursor exposure time of 0.4 s was chosen as a parameter for the preparation of the SiO₂ layer in the UV AR film because it facilitates the thickness uniformity [40].

3.3. Properties of the HfO₂ films of group G3

The ellipsometrically determined thicknesses of HfO₂ films grown at 50, 100, 150, and 200 °C are 72.4 nm, 57.3 nm, 51.5 nm, and 47.7 nm, respectively. The XRD spectra indicate that the HfO₂ films tend to crystallize with increasing temperature, as shown in Fig. 6a. Fig. 6b shows that the O/Hf ratio of the HfO₂ film grown at a temperature of 150 °C is the highest, and the scattering of the HfO₂ films grown at 150 and 200 °C is relatively lower than other temperatures. The chemical composition characterized by XPS indicates that Ar, C, and N elements are present in all HfO₂ films. Fig. 6c shows a typical XPS spectrum of Ar, C, and N elements in a HfO₂ film deposited at 200 °C. In addition to the two peaks at 241.9 eV (Ar 2p_{3/2}) and 243.9 eV (Ar 2p_{1/2}), which are inconsistent with that observed in SiO₂ film, another peak at 238.8 eV is observed in the Ar 2p spectrum corresponding to N KLL Auger [42]. Two noticeable peaks—at 285.0 eV and 289.0 eV—are observed in the C 1s spectrum, corresponding to the C–H bond and O=C=O bond, respectively [44]. In the core level spectrum of N 1s, in addition to the peak at 399.1 eV associated with the undoped amine unit, a peak at 395.8 eV associated with the Hf–N bond from Hf precursor is observed [45,49]. A typical Hf 4f spectrum of a HfO₂ film deposited at a temperature of 200 °C is shown in Fig. 7a and can be fitted into two groups of Hf 4f peaks. Each Hf 4f peak shows a sharp doublet according to spin-orbit splitting into the Hf 4f_{5/2} and Hf 4f_{7/2} with the splitting binding energy of ~1.7 eV in good agreement with the theoretical value. The binding energy of the Hf 4f_{7/2} peak located at 16.8 eV represents the Hf–O bond of HfO₂ [50]. The lower binding energy of the Hf 4f_{7/2} peak located at 15.9 eV represents the Hf–N bond [51], which indicates the presence of residual TEMAH precursor in HfO₂ film. A typical O 1s spectrum of a HfO₂ film deposited at a temperature of 200 °C is shown in Fig. 7b and can be fitted into two Gaussian peaks, centered at 530.3 eV (O–Hf) and 532.1 eV (Hf–O–C) [44], respectively. The presence of the O 1s peak at 532.1 eV and C 1s peak at 289 eV may indicate that precursor residues in the HfO₂ film were further oxidized by plasma oxygen. The individual and total contents of Ar, C, and N in the HfO₂ films are shown in Fig. 6d. Overall, the impurity contents in HfO₂ films are quite high, and the total impurity content in some HfO₂ films is even higher than 10%. The total impurity content of the HfO₂ film first increases with the increase of the deposition temperature, which may be due to the enhancement of chemisorption caused by the increase of chemical activity of the precursor [52]. When the deposition temperature is higher than 150 °C, the precursor tends to decompose due to the limited thermal stability [53], and the impurity content accordingly decreases with the deposition temperature. All HfO₂ films exhibit a large absorption as shown in Fig. 6e. The absorption of HfO₂ films decreases with measurement time and becomes stable after approximately 1500 s. This absorption annealing effect of HfO₂ films is possibly due to the interaction of 355-nm laser radiation with surface defects [54]. When the deposition temperature is lower than 150 °C, the film absorption is mainly caused by impurities, and increases with increasing deposition temperature. Although the HfO₂ film grown

at 200 °C shows a lower impurity content than the HfO₂ film grown at 150 °C, it shows a larger absorption, which may be due to lower reactive-site (such as hydroxyl) density [55] and therefore more oxygen vacancies caused by incomplete oxidation at high temperature [56]. The laser damage probability of the HfO₂ films is compared in Fig. 6f. The LIDT of HfO₂ films decreases with increasing temperature, which corresponds to the higher impurity content and lower O/Hf ratio at high temperatures.

3.4. 355 nm AR films for high power lasers

A HfO₂/SiO₂ double-layer AR film was designed to achieve a reflectance <0.2% at 355 nm. The growth temperature of the PEALD AR film was selected to be 150 °C for two reasons. First, the SiO₂ layer is on top of the HfO₂ layer and will be directly exposed to laser

radiation. Secondly, the thickness of HfO₂ is much thinner than that of SiO₂. The temperature that is beneficial to obtain a SiO₂ film with a higher LIDT is more favorable for obtaining an AR film with a higher LIDT.

As shown in Fig. 8a, the measured reflectance of both PEALD and e-beam AR films is lower than 0.2% as designed. The absorption curves versus measurement time of the PEALD and e-beam deposited AR films are shown in Fig. 8b. The absorption of the PEALD AR film is lower than the e-beam deposited AR film. The decreasing absorption of both PEALD and e-beam AR films versus measurement time is mostly due to the annealing effect of the underlying HfO₂ layer. The laser damage probability of the PEALD and e-beam deposited AR films as a function of the input fluence is compared in Fig. 8c. The LIDT of the PEALD AR film (24.4 J/cm²) is higher than that of the e-beam AR film (20.6 J/cm²), which

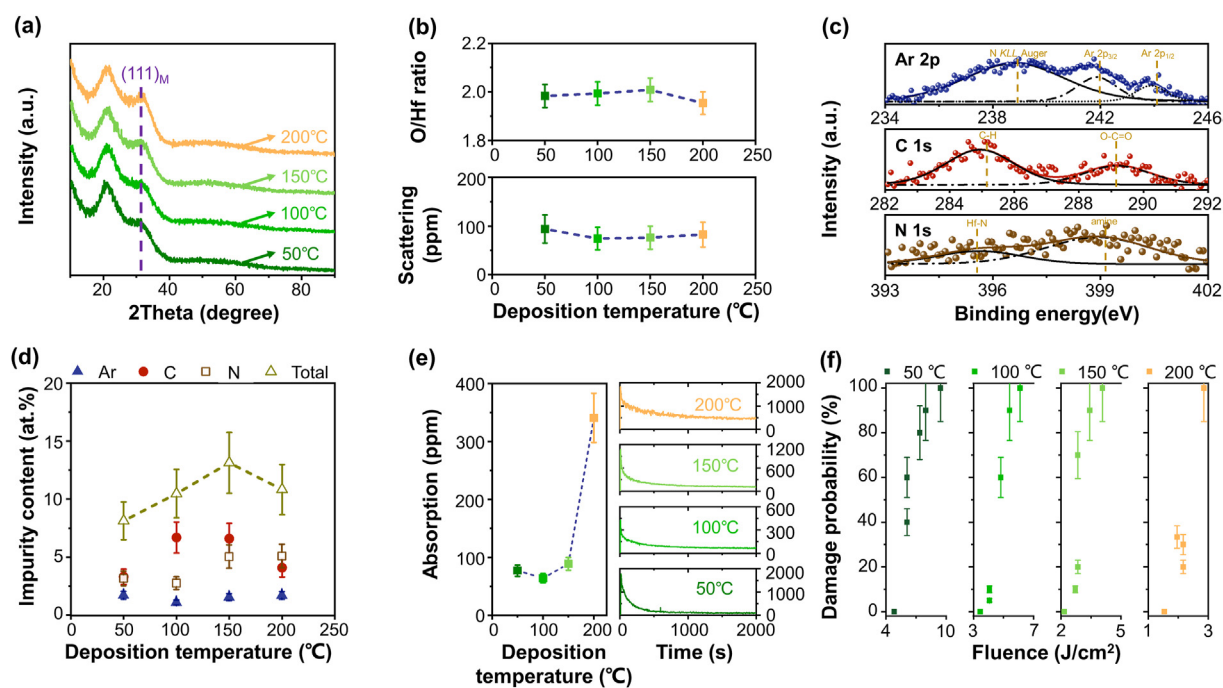


Fig. 6. (a) XRD spectra, (b) O/Hf ratio, and scattering measured at 632.8 nm of the PEALD HfO₂ films in Group G3. (c) A typical XPS spectrum of Ar 2p, C 1s, and N 1s in a PEALD HfO₂ film grown at 200 °C. (d) Individual and total impurity contents of Ar, C, and N elements, (e) absorption measured at 355 nm, (f) laser-induced damage probability for PEALD HfO₂ films in Group G3.

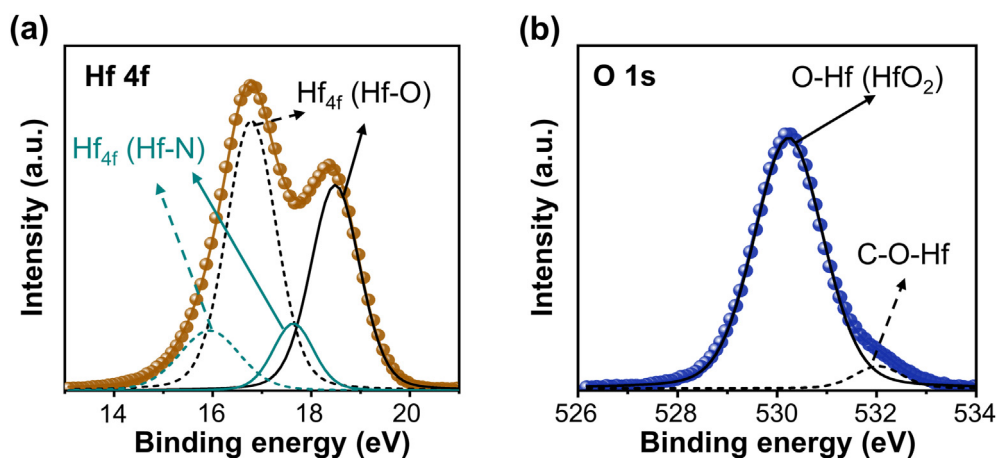


Fig. 7. The typical (a) Hf 4f spectrum and (b) O 1s spectrum of a PEALD HfO₂ film deposited at a temperature of 200 °C.

corresponds to the lower absorption of the PEALD AR film. The laser-induced damage probability of the fused-silica substrate with the same cleaning procedure as AR film is also shown in Fig. 8c for comparison.

The laser-induced damage morphology of PEALD AR film was characterized by FIB-SEM to investigate the laser-induced damage mechanism of PEALD AR film. As shown in Fig. 8, the typical damage morphology of a PEALD AR film is different from the observed pit damage morphology in a PEALD SiO₂ film and the ablation damage morphology in a PEALD HfO₂ film [40]. As shown in Fig. 8d, the damage morphology induced by a near-LIDT laser fluence indicates the formation of a blister and the buckling is initially located at the interface of HfO₂ and SiO₂ layers, which may be due to the high interface defect density [6,40], poor interfacial adhesion [57], and the stress mismatch of HfO₂ and SiO₂ layers [58,59]. At a higher laser fluence, the top SiO₂ layer has been stripped away (Fig. 8e), and no Hf element was observed in the EDS scan of the center area of the damaged site (Fig. 9), suggesting that the underlying HfO₂ layer is completely ablated at a higher laser fluence. As the laser fluence further increases, laser damage will penetrate the fused silica substrate, and a shell-like pit similar to the damage morphology of fused silica is observed in the center of the damaged area (Fig. 8f).

The damage morphology suggests that the initial damage of the PEALD AR film is closely related to the HfO₂ layer. Measures that can reduce impurity content and film absorption of the HfO₂ layer, including post-annealing [60] and incorporation of Al₂O₃ [61], may further improve the LIDT performance.

4. Conclusion

PEALD SiO₂ and HfO₂ films were prepared, and the properties related to UV laser applications were studied. Through a comprehensive analysis of the chemical composition, absorption, scattering and laser-induced damage property of PEALD SiO₂ and HfO₂ films, we confirmed that PEALD SiO₂ films have lower impurity content and absorption, and therefore show higher LIDT than PEALD HfO₂ films. This makes PEALD SiO₂ films suitable for UV laser applications. A bilayer structure HfO₂-SiO₂ AR film for 355 nm lasers was designed and fabricated via PEALD and e-beam deposition. The reflectance and LIDT of the PEALD AR films were investigated and compared with e-beam AR films. The reflectance of both AR films is lower than 0.2% as designed. The LIDT of the PEALD AR film (24.4 J/cm²) is higher than that of the e-beam AR film (20.6 J/cm²). The laser-induced damage morphology suggests that the laser-induced damage of PEALD AR film is closely related to the HfO₂

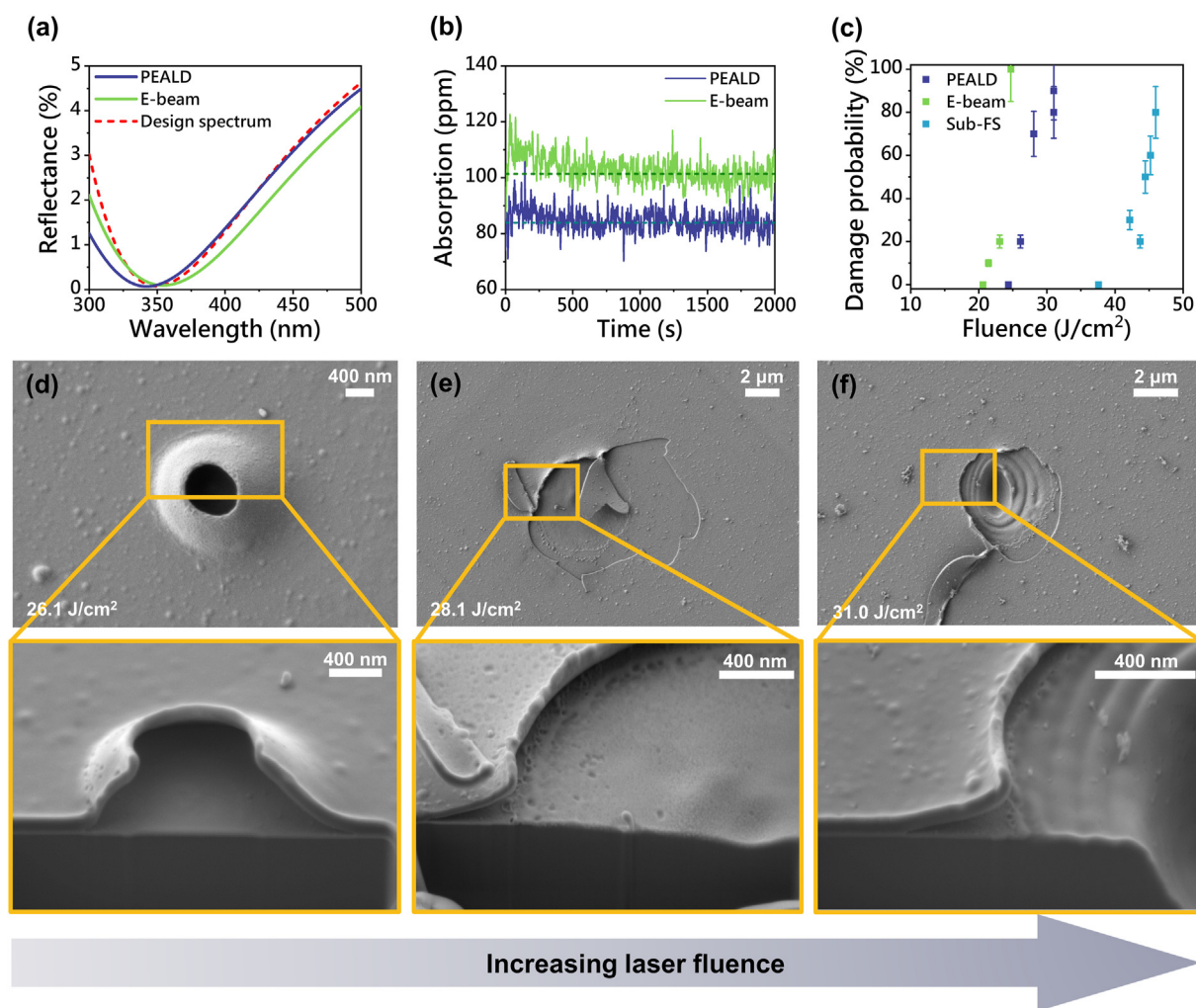


Fig. 8. (a) Measured reflectance spectra, (b) absorption curves versus measurement time, and (c) laser-induced damage probability of the PEALD and e-beam AR films and fused-silica substrate. The dash-dot line in (b) shows the average absorption between 1900 s and 2000 s of the measurement time. (d-f) Damaged sites imaged by SEM and the depth profiles of the marked regions measured by FIB.

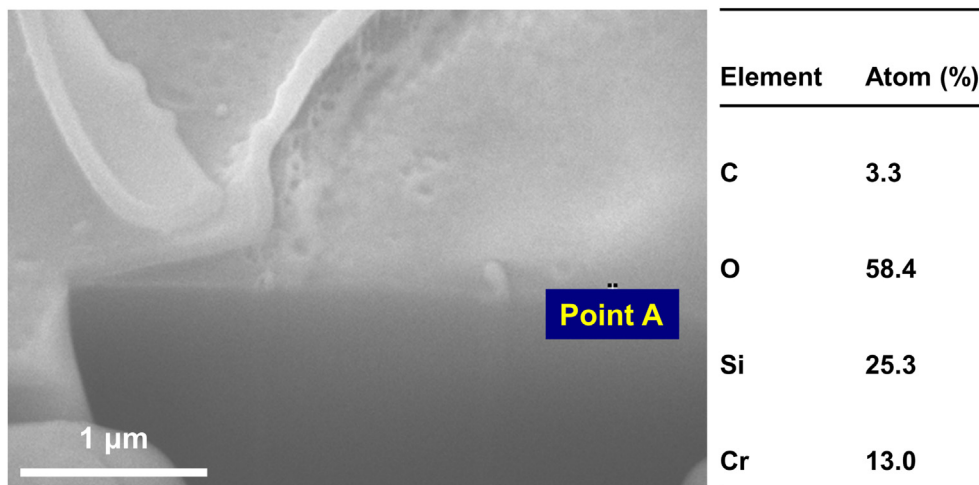


Fig. 9. EDS characterized element content of point 'A' in the center area of the damaged site of a PEALD AR film irradiated by a laser fluence of 28.1 J/cm².

layer. Further improvement of LIDT may be achieved by Al₂O₃ incorporation and post-annealing. PEALD-grown HfO₂-SiO₂ film is expected to be promising in AR films for laser applications.

CRedit authorship contribution statement

Chaoyi Yin: Conceptualization, Methodology, Formal analysis, Data curation, Writing - original draft, Visualization. **Meiping Zhu:** Validation, Writing - review & editing, Supervision, Project administration, Funding acquisition. **Tingting Zeng:** Validation, Formal analysis, Data curation, Writing - review & editing. **Chen Song:** Investigation, Resources. **Yingjie Chai:** Software, Formal analysis, Writing - review & editing. **Yuchuan Shao:** Validation, Project administration. **Rongjun Zhang:** Validation, Supervision. **Jiaoling Zhao:** Resources, Writing - review & editing. **Dawei Li:** Investigation, Supervision. **Jianda Shao:** Project administration, Funding acquisition.

Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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