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## Lanthanum Sulfide powder Analyzed by XPS

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X-ray photoelectron spectroscopy was performed on as-received, gamma lanthanum sulfide ( $\gamma$ -La<sub>2</sub>S<sub>3</sub>) powder (99.9%), doped with sodium sulfide (Na<sub>2</sub>S) 0.1% wt%. The scans provide photoelectron spectroscopy investigation data for cubic La<sub>2</sub>S<sub>3</sub> to help with identification of lanthanum compounds. This report includes charge corrected scans for the survey along with S 2s, S 2p, La 4s, La 4p, La 4d, La 3p<sub>3/2</sub>, La 3p<sub>1/2</sub>, La 3d, Na 1s, O 1s, and C 1s surface photoelectron signals.

**Keywords:** La<sub>2</sub>S<sub>3</sub>; Lanthanides; transition metal dichalcogenide; x-ray photoelectron spectroscopy; XPS

### INTRODUCTION

Lanthanum sulfide (La<sub>2</sub>S<sub>3</sub>) is a commodity chemical for production of arsenic-free, chalcogenide glasses and ceramics[1-4] that transmit infrared radiation. For example, La<sub>2</sub>S<sub>3</sub> is a major precursor in the production of gallium lanthanum sulfide and strontium lanthanum sulfide, both of which are chalcogenides.[4-7] La<sub>2</sub>S<sub>3</sub>-based glasses have unique characteristics such as, high refractive index (2.41n ~ 0.01),[1, 8, 9] and an optical bandgap of ~2.6 eV. [5, 8, 10] In the glassy phase, the high content of lanthanum results in a high dispersion of ions in the matrix giving greater ionic conductivity (1.01x10<sup>-3</sup> S/cm).[6, 11] Optically, these glasses have high transmission from the visible wavelength (380-750 nm) to infrared (~ 8  $\mu$ m).[1, 4, 12] However, there are no reports on the x-ray photoelectron spectroscopy (XPS) of La<sub>2</sub>S<sub>3</sub>.

In this study, we analyze  $\gamma$ -La<sub>2</sub>S<sub>3</sub> by XPS to study and quantify the bonding states of observable electron shells for La and S, along with minor species Na, C, and O. Sodium sulfide has been introduced at a 0.1 wt% level during the synthesis of the compound to stabilize the cubic phase of La<sub>2</sub>S<sub>3</sub> at room temperature. [13]. The use of XPS also provides quantification of atomic percentages (*i.e.*, stoichiometry) along with the bonding states of additional constituent elements and non-trace level impurities, if present. From the lanthanum fine spectra La 3d along with the O 1s spectra we see that the material is not a pure metal sulfide but a mixture of Lanthanum Oxide and Lanthanum Sulfide. The shake-up peaks in the La 3d along with the O 1s metal oxide peak resemble the feature from M. F. Sunding et al.[14] study on lanthanum oxide. There are some notable differences to peak height and shape but do give insights to the material not being pure  $\gamma$ -La<sub>2</sub>S<sub>3</sub>. While looking at the Sulfur fine

**Accession#:** 01808

**Technique:** XPS

**Host Material:**  $\gamma$ -La<sub>2</sub>S<sub>3</sub>, powder doped with Na<sub>2</sub>S

**Instrument:** XPS ESCALAB 250Xi Fisher Scientific

**Major Elements in Spectra:** La, S

**Minor Elements in Spectra:** O, C, Na

**Published Spectra:** 12

**Spectral Category:** Comparison

spectra there is no evidence of oxidized sulfur that would be contributing to the O 1s peak.

### SPECIMEN DESCRIPTION (ACCESSION # 01808)

**Host Material:** Lanthanum Sulfide  $\gamma$ -La<sub>2</sub>S<sub>3</sub> Bulk Powder

**CAS Registry #:** 12031-49-1

**Host Material Characteristics:** homogeneous; powder; polycrystalline; semiconductor; inorganic compound; Powder

**Chemical Name:** Lanthanum Sulfide 99.9% wt% doped with Sodium Sulfide 0.1% wt%

**Source:** Lorad Chemical Corporation 1200 19<sup>th</sup> Street North St. Petersburg, FL 33713 USA

**Host Composition:** La<sub>2</sub>S<sub>3</sub> 99.9% wt% & Na<sub>2</sub>S 0.1% wt%

**Form:** 99.9% purity powder gamma phase La<sub>2</sub>S<sub>3</sub>

**Structure:** Modified Th<sub>3</sub>P<sub>4</sub>

**History & Significance:** The specimen was received in a vacuum sealed glass bottle and stored in an inert atmosphere chamber at room temperature. La<sub>2</sub>S<sub>3</sub> is commonly used in glasses, ceramics, and optical fibers.

**As Received Condition:** Light yellow powder

**Analyzed Region:** Same as host material

**Ex Situ Preparation/Mounting:** Powder mounted on double-sides carbon tape.

**In Situ Preparation:** 120 second argon ion sputtering was used to clean the surface before analysis

**Charge Control:** Charge compensation is delivered by both an in-lens electrostatic electron flood source (1 eV, 100  $\mu$ A) and a dual-beam low energy electron and ion coaxial flood source (2 eV, 100  $\mu$ A).

**Temp. During Analysis:** 300K

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**Pressure During Analysis:**  $5 \times 10^{-8}$  Pa

**Pre-analysis Beam Exposure:** 0 s

#### INSTRUMENT DESCRIPTION

**Manufacturer and Model:** Thermo Fisher Scientific ESCALAB 250Xi

**Analyzer Type:** spherical sector

**Detector:** Channeltron

**Number of Detector Elements:** 6

#### INSTRUMENT PARAMETERS COMMON TO ALL SPECTRA

##### ■ Spectrometer

**Analyzer Mode:** constant pass energy

**Throughput (T=EN<sup>2</sup>):** Calculated from a polynomial fit to a plot of log [peak area/(PE x XSF)] versus log[KE/PE], where PE is the pass energy, KE is the kinetic energy, and XSF is the relative sensitivity factor.

**Excitation Source Window:** None

**Excitation Source:** Al  $K_{\alpha}$  monochromatic

**Source Energy:** 1486.6 eV

**Source Strength:** 200 W

**Source Beam Size:** 200  $\mu\text{m}$  x 200  $\mu\text{m}$

**Signal Mode:** single channel direct

##### ■ Geometry

**Incident Angle:** 58 °

**Source-to-Analyzer Angle:** 58 °

**Emission Angle:** 0 °

**Specimen Azimuthal Angle:** 90 °

**Acceptance Angle from Analyzer Axis:** 45 °

**Analyzer Angular Acceptance Width:** 22.5 ° x 22.5 °

##### ■ Ion Gun

**Manufacturer and Model:** Thermo Fisher Scientific EX03 Ion Gun System

**Energy:** 3000 eV

**Current:** 0.02 mA

**Current Measurement Method:** biased stage

**Sputtering Species:** Ar<sup>+</sup>

**Spot Size (unrastered):** 500  $\mu\text{m}$

**Raster Size:** 4500  $\mu\text{m}$  x 4500  $\mu\text{m}$

**Incident Angle:** 40 °

**Polar Angle:** 40 °

**Azimuthal Angle:** 270 °

**Comment:** These parameters correspond to ion cleaning methods used in typical operation requiring surface cleaning

#### DATA ANALYSIS METHOD

**Energy Scale Correction:** binding energy scale was reference to O 1s = 531 eV

**Recommended Energy Scale Shift:** Shift +0.58 eV

**Peak Shape and Background Method:** Thermo Scientific Avantage software version 5.9902 was used for peak fitting and background subtraction. The smart (Shirley function) was used to subtract the background for La 3d, S 2p, O 1s, C 1s, Na 1s peaks. Using the smart feature, utilizes constraints that limit the background from having greater intensity than data from points in the collection region.

**Quantitation Method:** Atomic percentages were calculated using the Thermo Scientific Avantage software version 5.9902. Along with atomic percentages Thermo scientific Avantage software sensitivity factors. The peak library is ALWAG [15]

#### ACKNOWLEDGMENTS

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#### AUTHOR DECLARATION

**Conflict of Interest:** The authors have no conflicts to disclose.

#### DATA AVAILABILITY STATEMENT

The data that supports the findings of this study are available within the article and its supplementary material.

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SPECTRAL FEATURES TABLE

Spectrum ID #	Element/Transition	Peak Energy (eV)	Peak Width FWHM (eV)	Peak Area (eV x cts/s)	Sensitivity Factor	Concentration (at. %)	Peak Assignment
01808-02	S 2s	224.62	2.78	8974.73	1.294	...	La <sub>2</sub> S <sub>3</sub>
01808-03	S 2p	161	2.58	12818.31	1.881	39.44	La <sub>2</sub> S <sub>3</sub>
01808-04 <sup>a</sup>	O 1s	528.14	1.33	13393.67	2.881	29.49	La <sub>2</sub> O <sub>3</sub> / Carbonates / Hydroxyls
01808-05	Na 1s	1071.23	1.85	4161.84	10.588	3.33	Na
01808-06	La 4s	274.09	6.3	4010.16	1.19	...	La <sub>2</sub> S <sub>3</sub> / La <sub>2</sub> O <sub>3</sub>
01808-07 <sup>a</sup>	La 4p	195.29	5.5	19642.48	4.33	...	La <sub>2</sub> S <sub>3</sub> / La <sub>2</sub> O <sub>3</sub>
01808-08 <sup>a</sup>	La 4d	104.27	5.95	51987.48	10.733	27.74	La <sub>2</sub> S <sub>3</sub> / La <sub>2</sub> O <sub>3</sub>
01808-09 <sup>a</sup>	La 3p <sub>3/2</sub>	1127.2	7.53	46502.55	14	...	La <sub>2</sub> S <sub>3</sub> / La <sub>2</sub> O <sub>3</sub>
01808-10 <sup>a</sup>	La 3p <sub>1/2</sub>	1206.72	7.26	18993.02	6.1	...	La <sub>2</sub> S <sub>3</sub> / La <sub>2</sub> O <sub>3</sub>
01808-11	La 3d <sub>5/2</sub>	837.11	2.28	196155.2	32.944	...	La <sub>2</sub> S <sub>3</sub> / La <sub>2</sub> O <sub>3</sub>
01808-12 <sup>b</sup>	C 1s	284.43	1	838.92	1	...	C-C/ Carbonates

<sup>a</sup> Peak assignment for La<sub>2</sub>O<sub>3</sub> sourced from M. F. Sunding et al.[14]

<sup>b</sup> Quantification of Carbon not included due to low confidence as high noise to signal ratio on fine spectra

ANALYZER CALIBRATION TABLE

Spectrum ID #	Element/Transition	Peak Energy (eV)	Peak Width FWHM (eV)	Peak Area (eV x cts/s)	Sensitivity Factor	Concentration (at. %)	Peak Assignment
...	Au 4f	84.05	0.57	262830.05	20.735	...	Au
...	Ag 3d	368.36	0.48	386600.57	22.131	...	Ag
...	Cu 2p	932.8	0.77	655133.11	26.513	...	Cu

GUIDE TO FIGURES

Spectrum (Accession) #	Spectral Region	Voltage Shift*	Multiplier	Baseline	Comment #
01808-01	Survey	-0.58	1	0	1
01808-02	S 2s	-0.58	1	0	1
01808-03	S 2p	-0.58	1	0	1
01808-04	O 1s	-0.58	1	0	1
01808-05	Na 1s	-0.58	1	0	1
01808-06	La 4s	-0.58	1	0	1
01808-07	La 4p	-0.58	1	0	1
01808-08	La 4d	-0.58	1	0	1
01808-09	La 3p <sub>3/2</sub>	-0.58	1	0	1
01808-10	La 3p <sub>1/2</sub>	-0.58	1	0	1
01808-11	La 3d	-0.58	1	0	1
01808-12	C 1s	-0.58	1	0	1

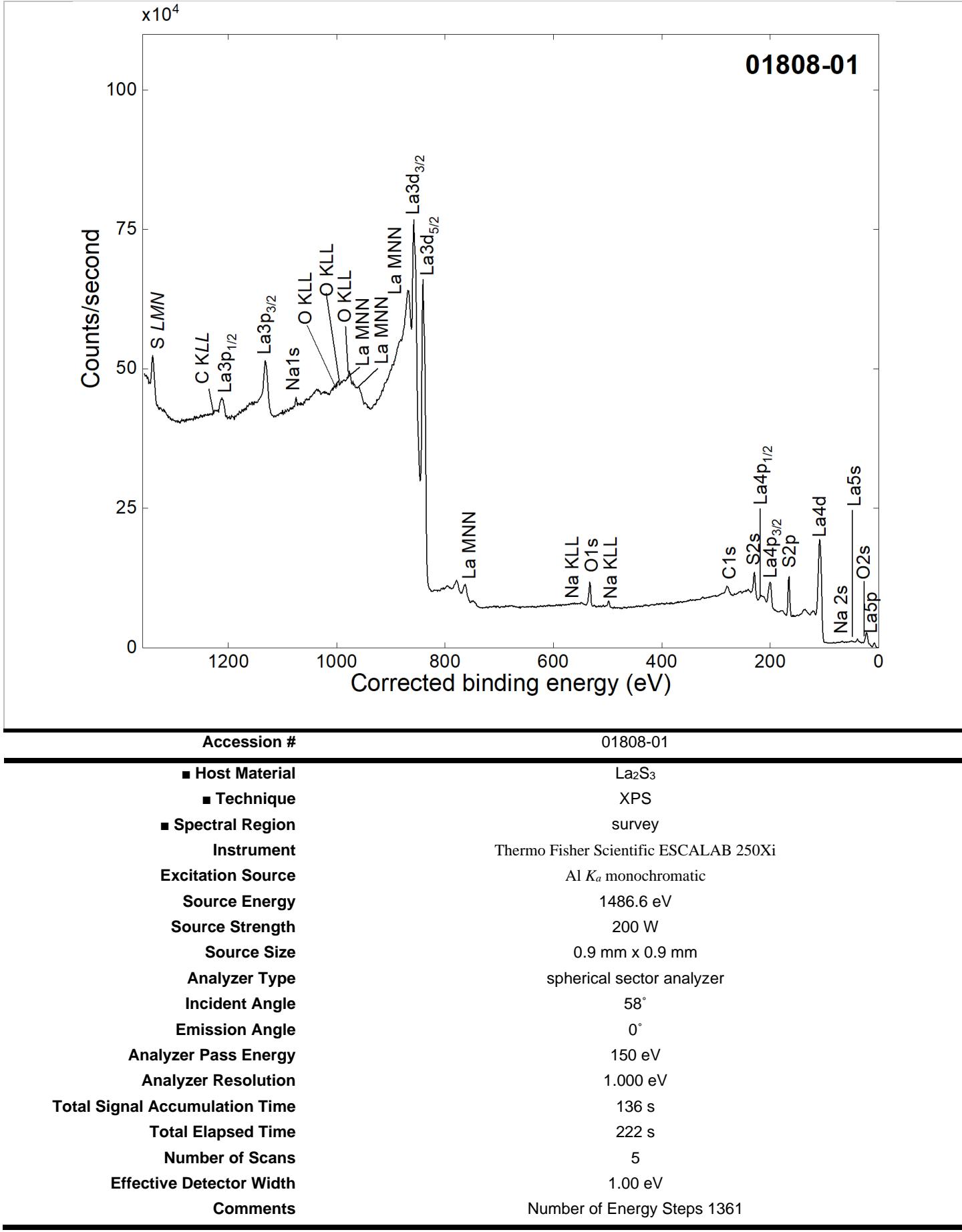
\*Voltage shift of the archived (as-measured) spectrum relative to the printed figure. The figure reflects the recommended energy scale correction due to a calibration correction, sample charging, flood gun, or other phenomenon.

1. La<sub>2</sub>S<sub>3</sub> powder



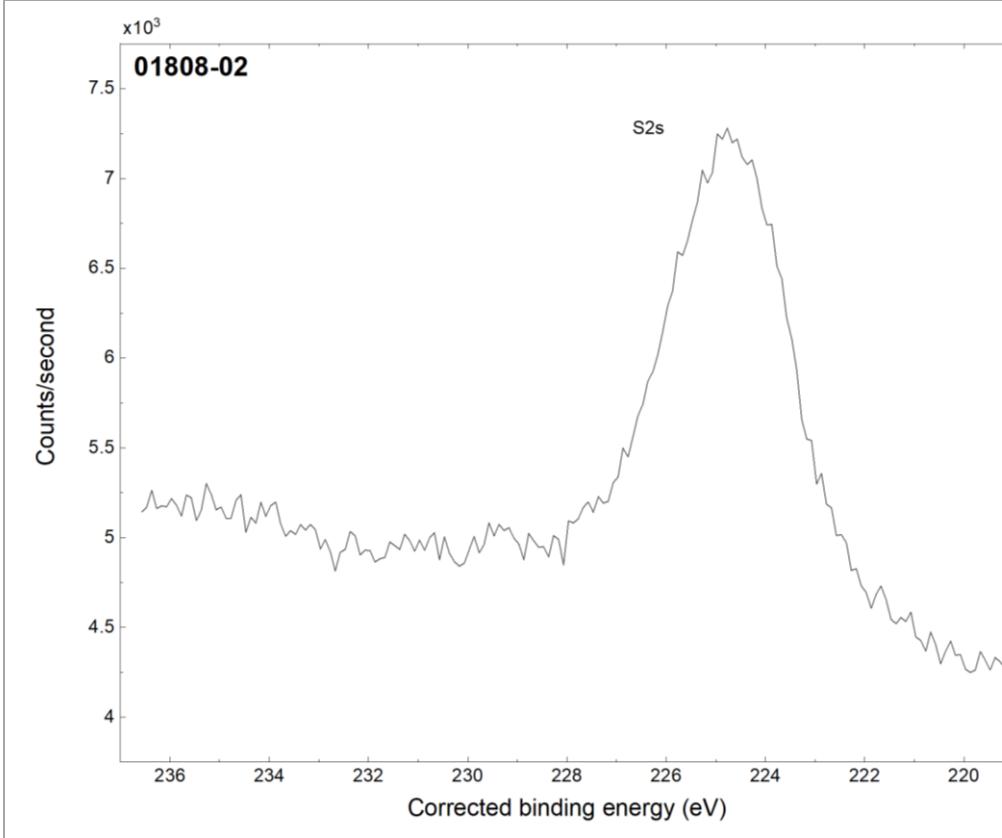


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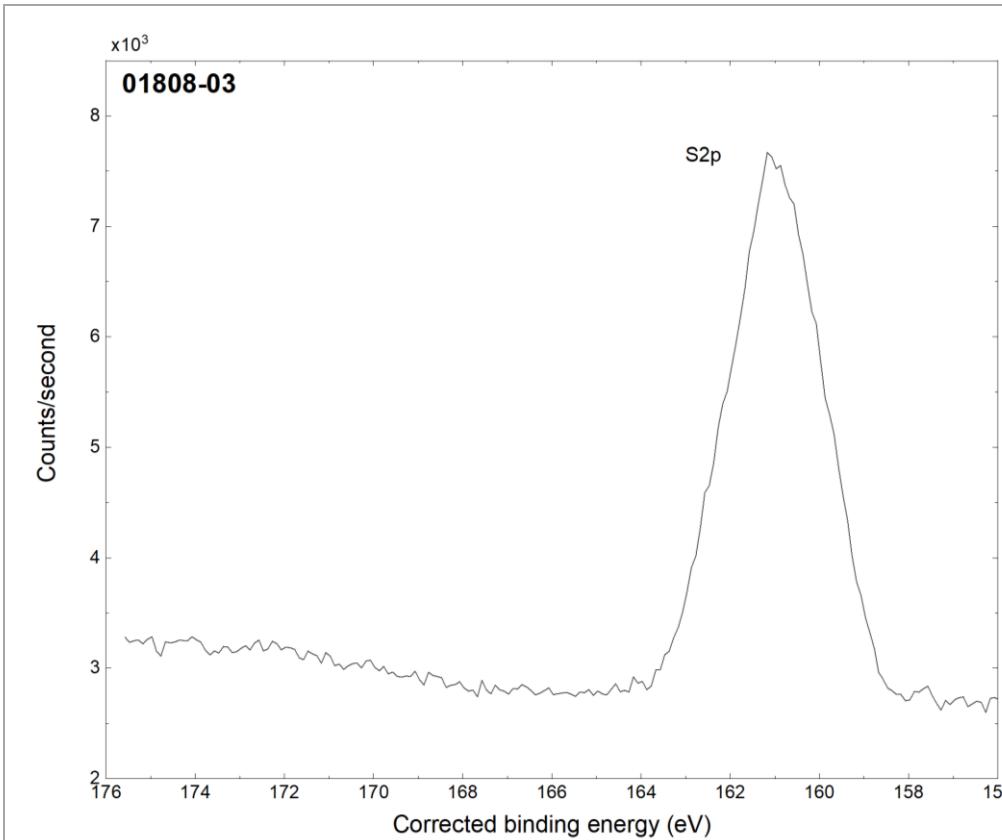




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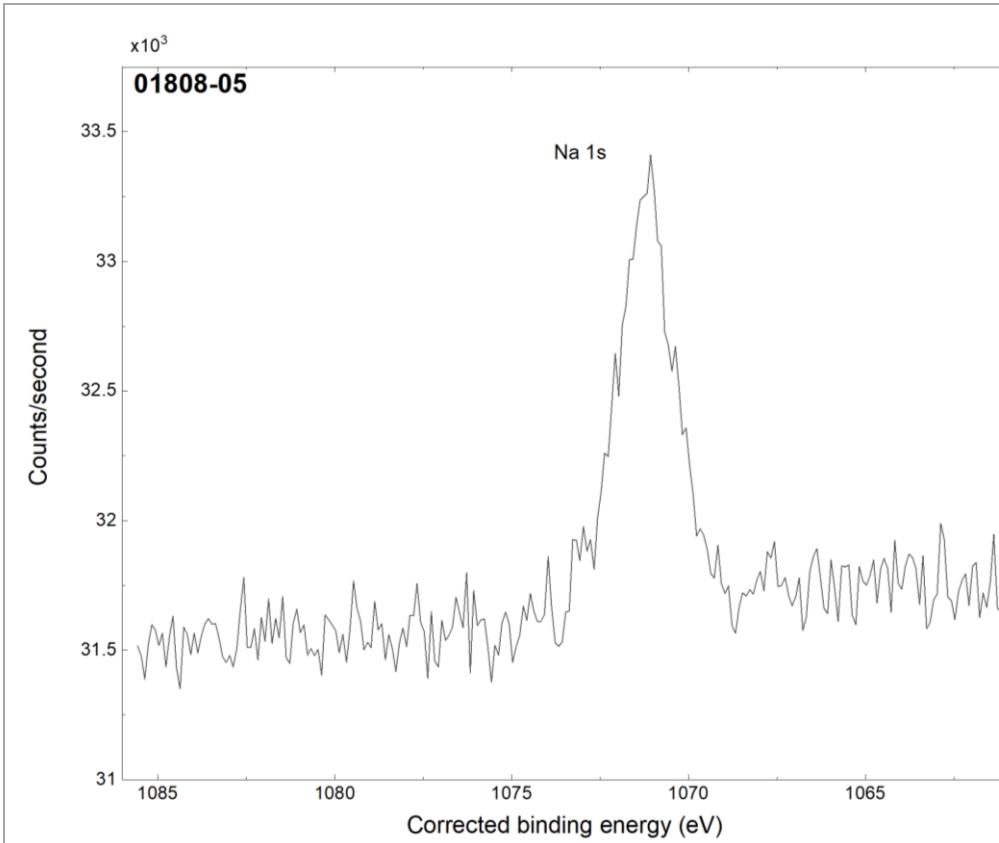
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■ **Spectral Region:** S 2s  
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Excitation Source:  $\text{Al K}_\alpha$  monochromatic  
Source Energy: 1486.6 eV  
Source Strength: 200 W  
Source Size: 0.9 mm x 0.9 mm  
Analyzer Type: spherical sector  
Incident Angle: 58 °  
Emission Angle: 0 °  
Analyzer Pass Energy 20 eV  
Analyzer Resolution: 0.100 eV  
Total Signal Accumulation Time: 180.9 s  
Total Elapsed Time: 230 s  
Number of Scans: 20  
Effective Detector Width: 0.100 eV



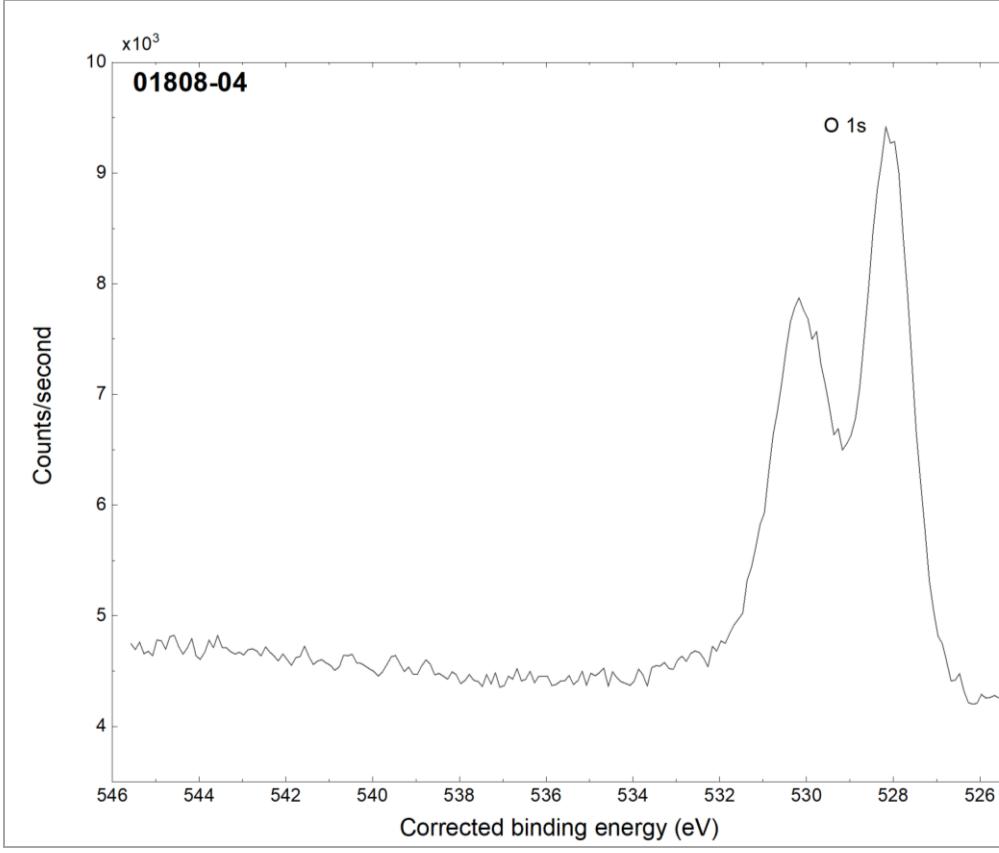
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■ **Technique:** XPS  
■ **Spectral Region:** S 2p  
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Excitation Source:  $\text{Al K}_\alpha$  monochromatic  
Source Energy: 1486.6 eV  
Source Strength: 200 W  
Source Size: 0.9 mm x 0.9 mm  
Analyzer Type: spherical sector  
Incident Angle: 58 °  
Emission Angle: 0 °  
Analyzer Pass Energy 20 eV  
Analyzer Resolution: 0.100 eV  
Total Signal Accumulation Time: 200.9 s  
Total Elapsed Time: 251 s  
Number of Scans: 20  
Effective Detector Width: 0.100 eV



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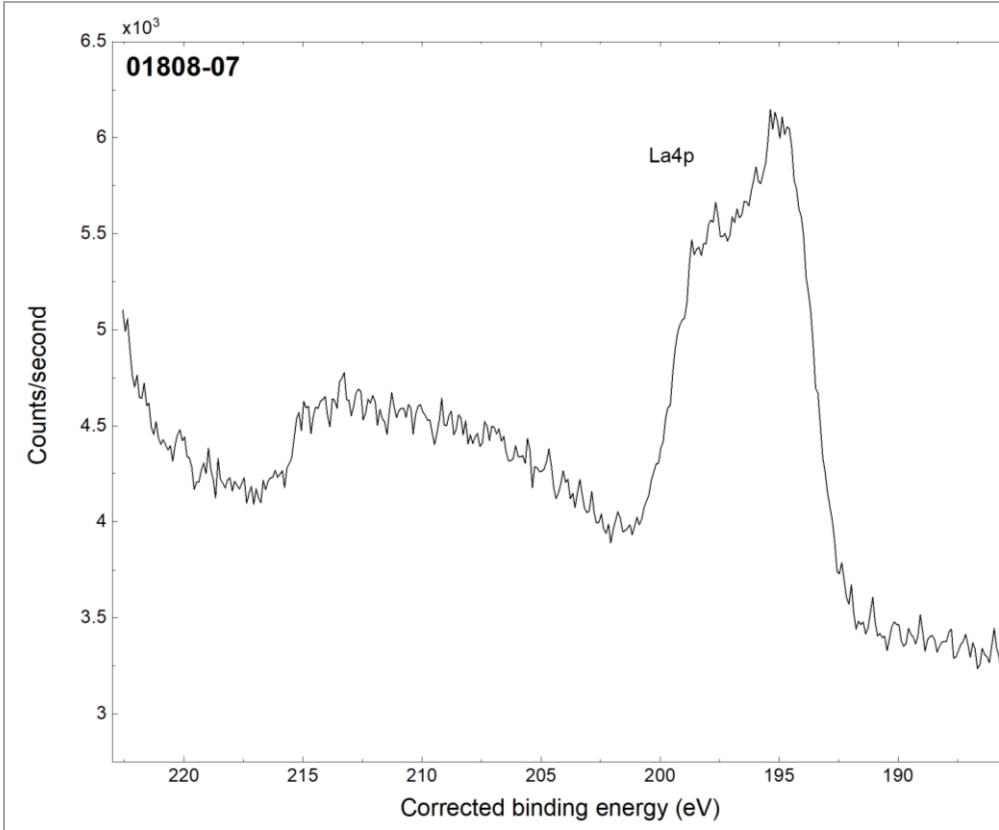
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■ **Spectral Region:** O 1s  
Instrument: Thermo Fisher Scientific ESCALAB 250Xi  
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Source Energy: 1486.6 eV  
Source Strength: 200 W  
Source Size: 0.9 mm x 0.9 mm  
Analyzer Type: spherical sector  
Incident Angle: 58 °  
Emission Angle: 0 °  
Analyzer Pass Energy 20 eV  
Analyzer Resolution: 0.100 eV  
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Total Elapsed Time: 253 s  
Number of Scans: 20  
Effective Detector Width: 0.100 eV



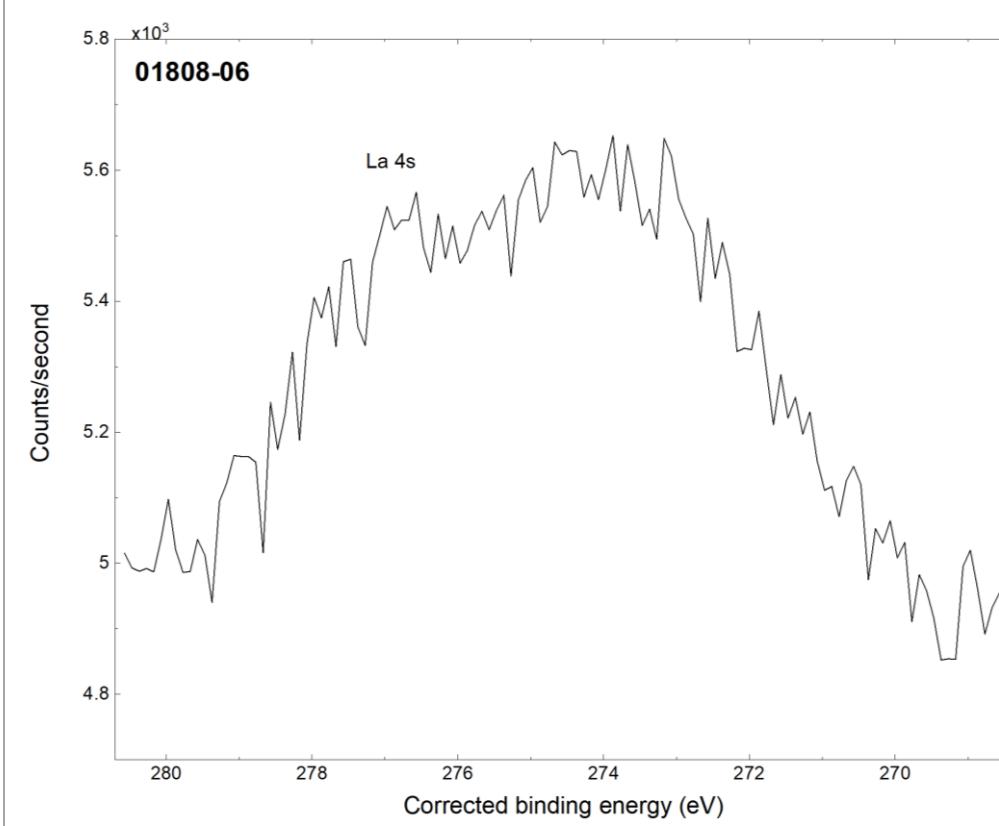
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■ **Technique:** XPS  
■ **Spectral Region:** Na 1s  
Instrument: Thermo Fisher Scientific ESCALAB 250Xi  
Excitation Source: Al  $K\alpha$  monochromatic  
Source Energy: 1486.6 eV  
Source Strength: 200 W  
Source Size: 0.9 mm x 0.9 mm  
Analyzer Type: spherical sector  
Incident Angle: 58 °  
Emission Angle: 0 °  
Analyzer Pass Energy 20 eV  
Analyzer Resolution: 0.100 eV  
Total Signal Accumulation Time: 627.2 s  
Total Elapsed Time: 762 s  
Number of Scans: 50  
Effective Detector Width: 0.100 eV



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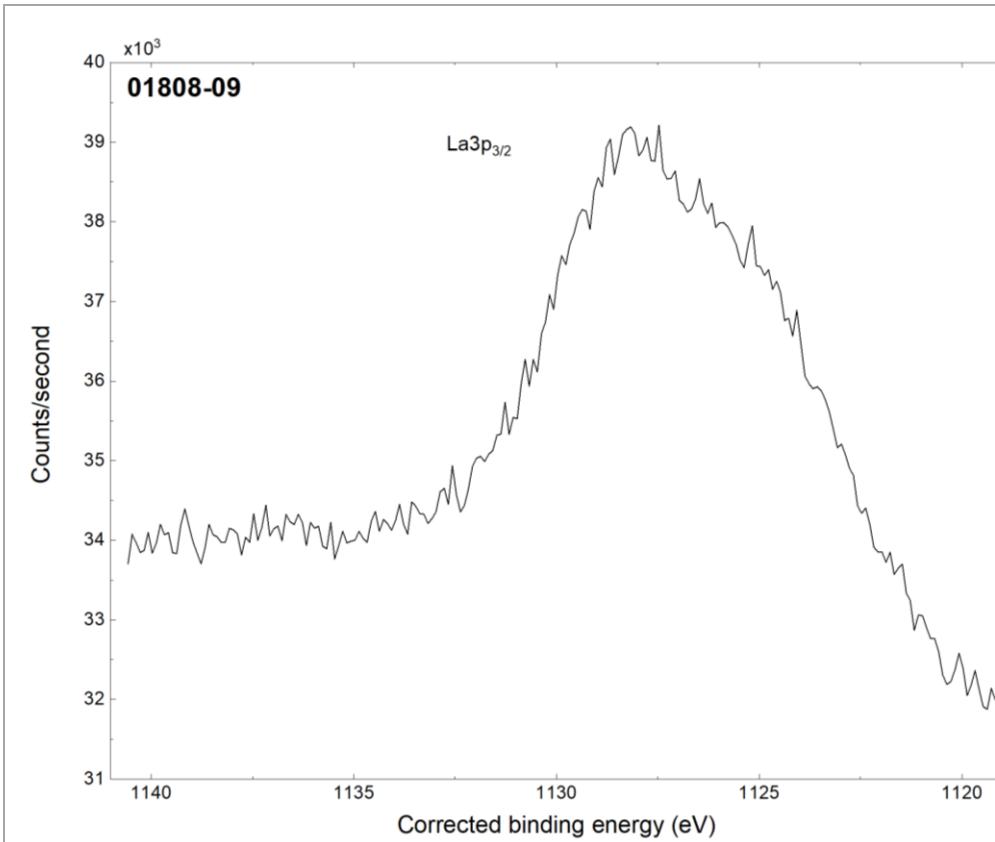
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■ Technique: XPS  
■ Spectral Region: La 4s  
Instrument: Thermo Fisher Scientific ESCALAB 250Xi  
Excitation Source: Al  $K_\alpha$  monochromatic  
Source Energy: 1486.6 eV  
Source Strength: 200 W  
Source Size: 0.9 mm x 0.9 mm  
Analyzer Type: spherical sector  
Incident Angle: 58 °  
Emission Angle: 0 °  
Analyzer Pass Energy 20 eV  
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Number of Scans: 20  
Effective Detector Width: 0.100 eV



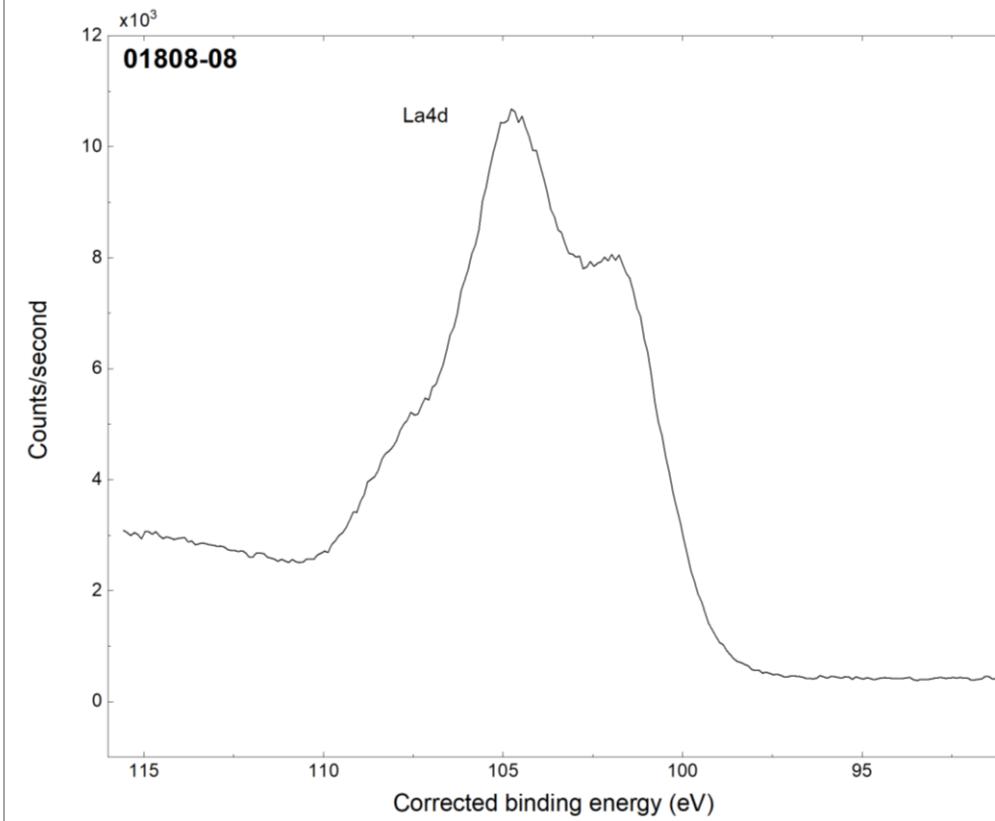
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Instrument: Thermo Fisher Scientific ESCALAB 250Xi  
Excitation Source: Al  $K_\alpha$  monochromatic  
Source Energy: 1486.6 eV  
Source Strength: 200 W  
Source Size: 0.9 mm x 0.9 mm  
Analyzer Type: spherical sector  
Incident Angle: 58 °  
Emission Angle: 0 °  
Analyzer Pass Energy 20 eV  
Analyzer Resolution: 0.100 eV  
Total Signal Accumulation Time: 370.9 s  
Total Elapsed Time: 432 s  
Number of Scans: 20  
Effective Detector Width: 0.100 eV



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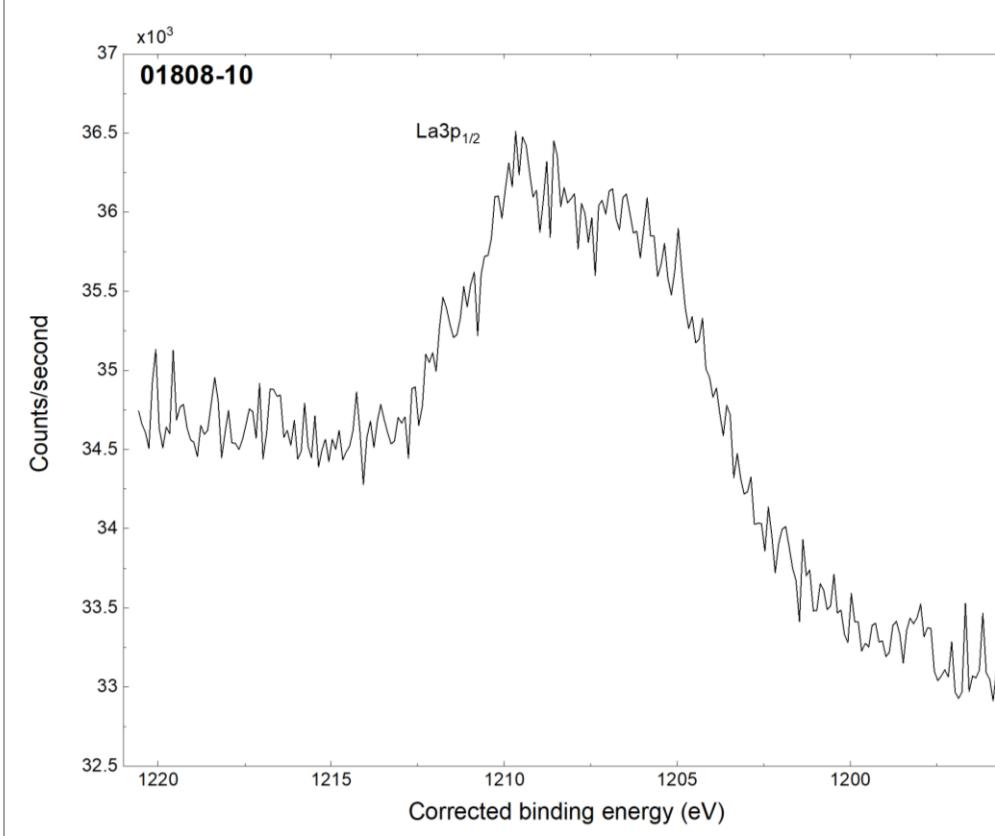
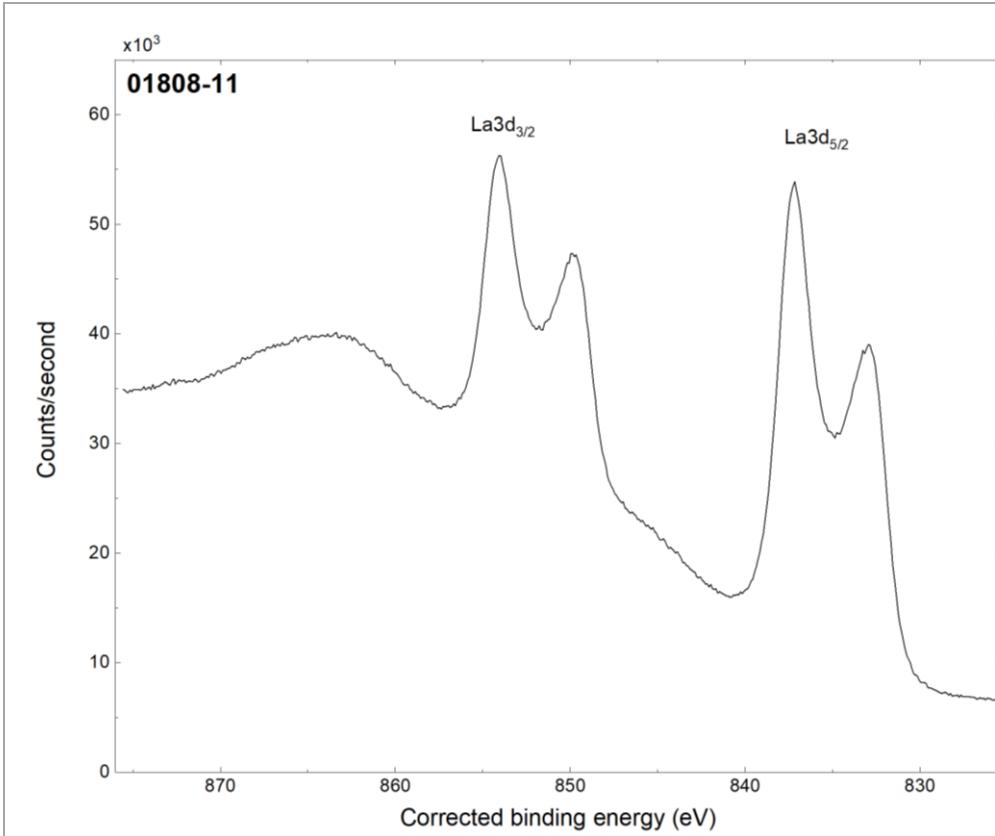
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■ **Technique:** XPS  
■ **Spectral Region:**  $\text{La}^{3p_{3/2}}$   
Instrument: Thermo Fisher Scientific ESCALAB 250Xi  
Excitation Source:  $\text{Al K}_\alpha$  monochromatic  
Source Energy: 1486.6 eV  
Source Strength: 200 W  
Source Size: 0.9 mm x 0.9 mm  
Analyzer Type: spherical sector  
Incident Angle: 58 °  
Emission Angle: 0 °  
Analyzer Pass Energy 20 eV  
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Number of Scans: 20  
Effective Detector Width: 0.100 eV



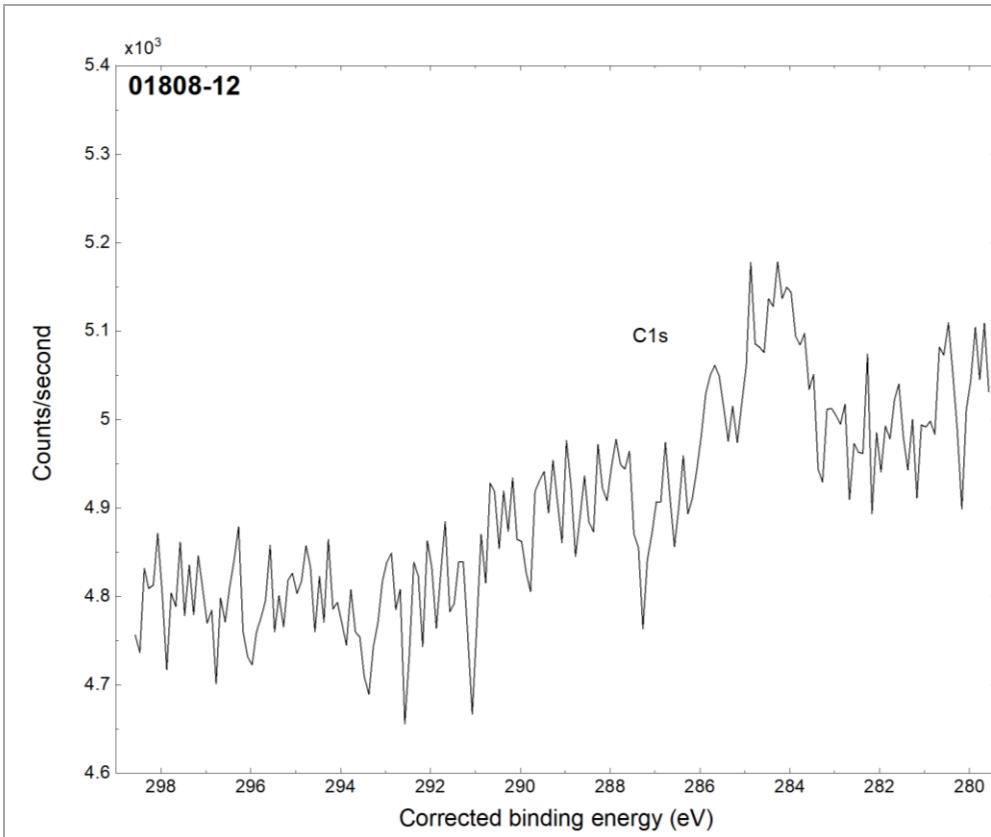
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Instrument: Thermo Fisher Scientific ESCALAB 250Xi  
Excitation Source:  $\text{Al K}_\alpha$  monochromatic  
Source Energy: 1486.6 eV  
Source Strength: 200 W  
Source Size: 0.9 mm x 0.9 mm  
Analyzer Type: spherical sector  
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Emission Angle: 0 °  
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Analyzer Resolution: 0.100 eV  
Total Signal Accumulation Time: 250.9 s  
Total Elapsed Time: 313 s  
Number of Scans: 20  
Effective Detector Width: 0.100 eV



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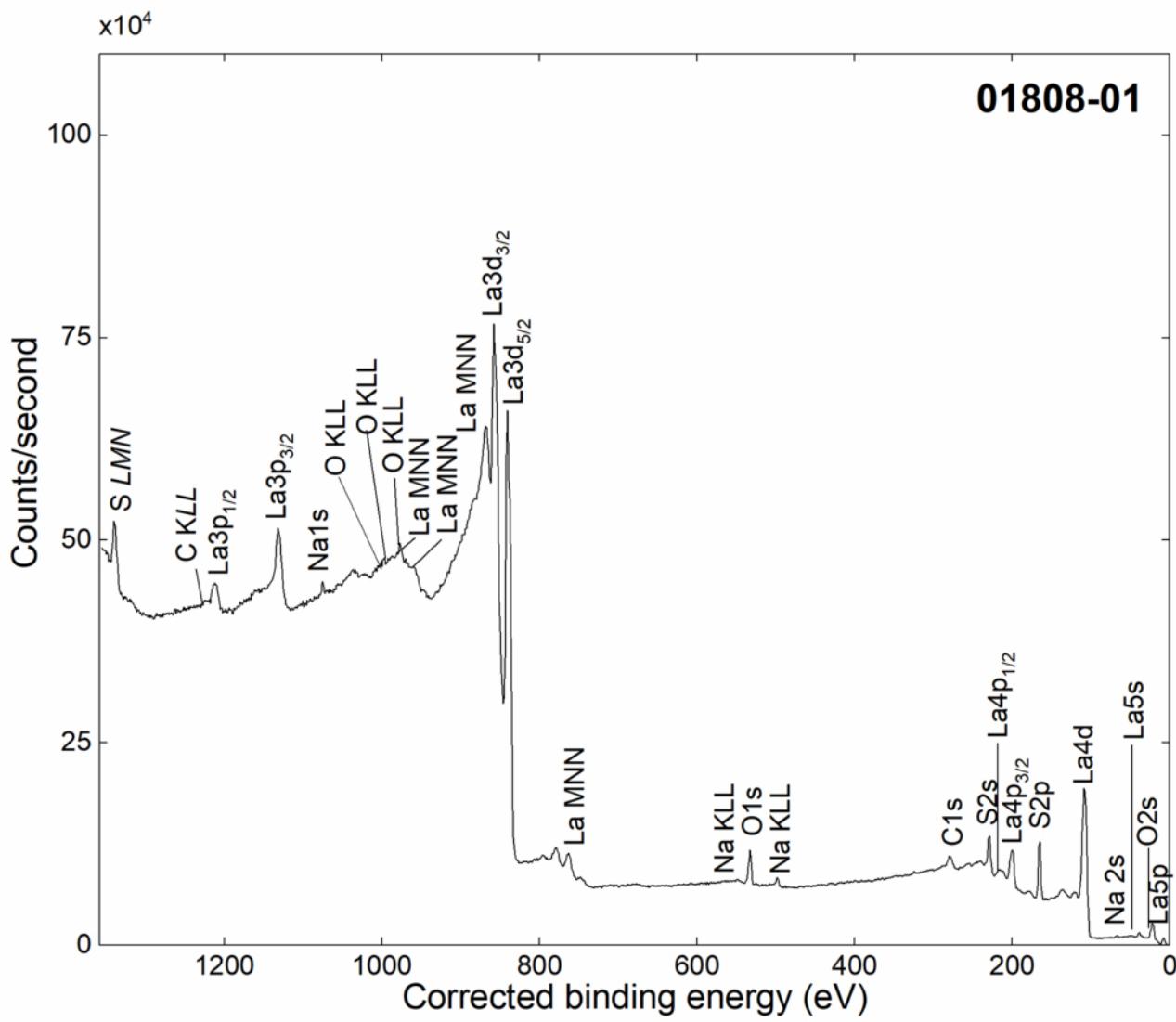
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■ **Accession #:** 01808-12  
■ **Host Material:** Lanthanum Sulfide  $\text{La}_2\text{S}_3$  Bulk Powder  
■ **Technique:** XPS  
■ **Spectral Region:** C 1s  
Instrument: Thermo Fisher Scientific ESCALAB 250Xi  
Excitation Source: Al  $K\alpha$  monochromatic  
Source Energy: 1486.6 eV  
Source Strength: 200 W  
Source Size: 0.9 mm x 0.9 mm  
Analyzer Type: spherical sector  
Incident Angle: 58 °  
Emission Angle: 0 °  
Analyzer Pass Energy 20 eV  
Analyzer Resolution: 0.100 eV  
Total Signal Accumulation Time: 238.7 s  
Total Elapsed Time: 301 s  
Number of Scans: 25  
Effective Detector Width: 0.100 eV

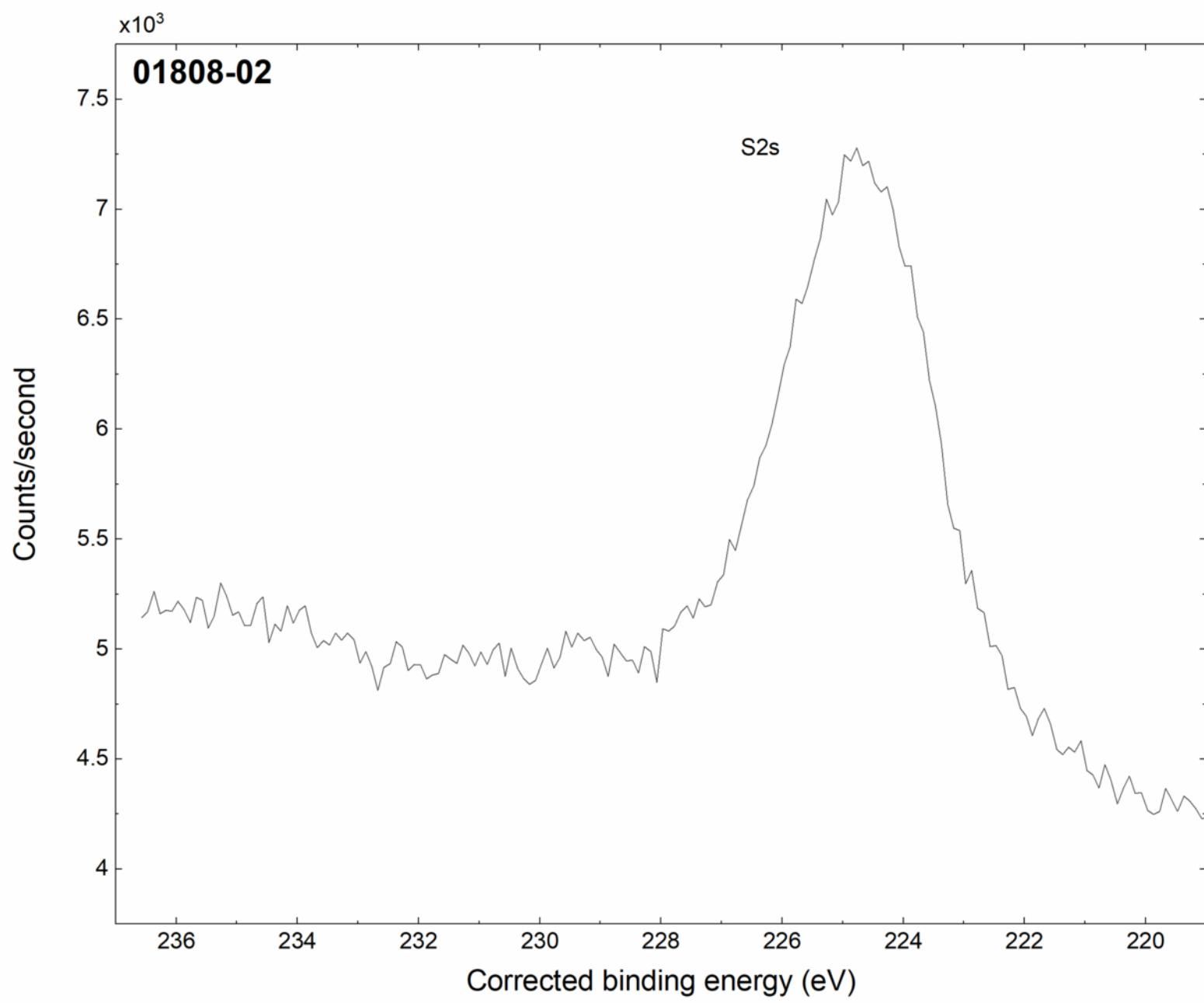


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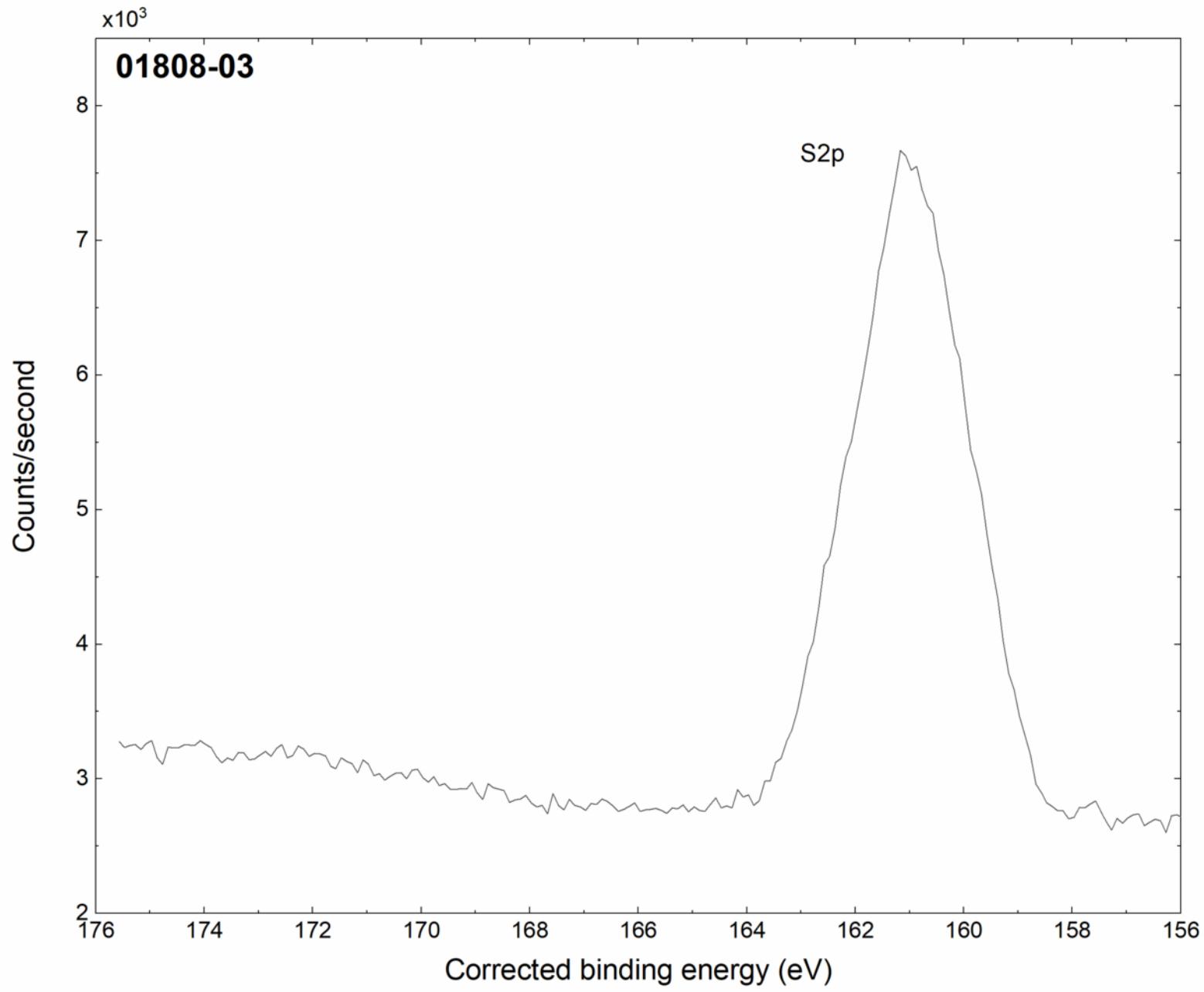


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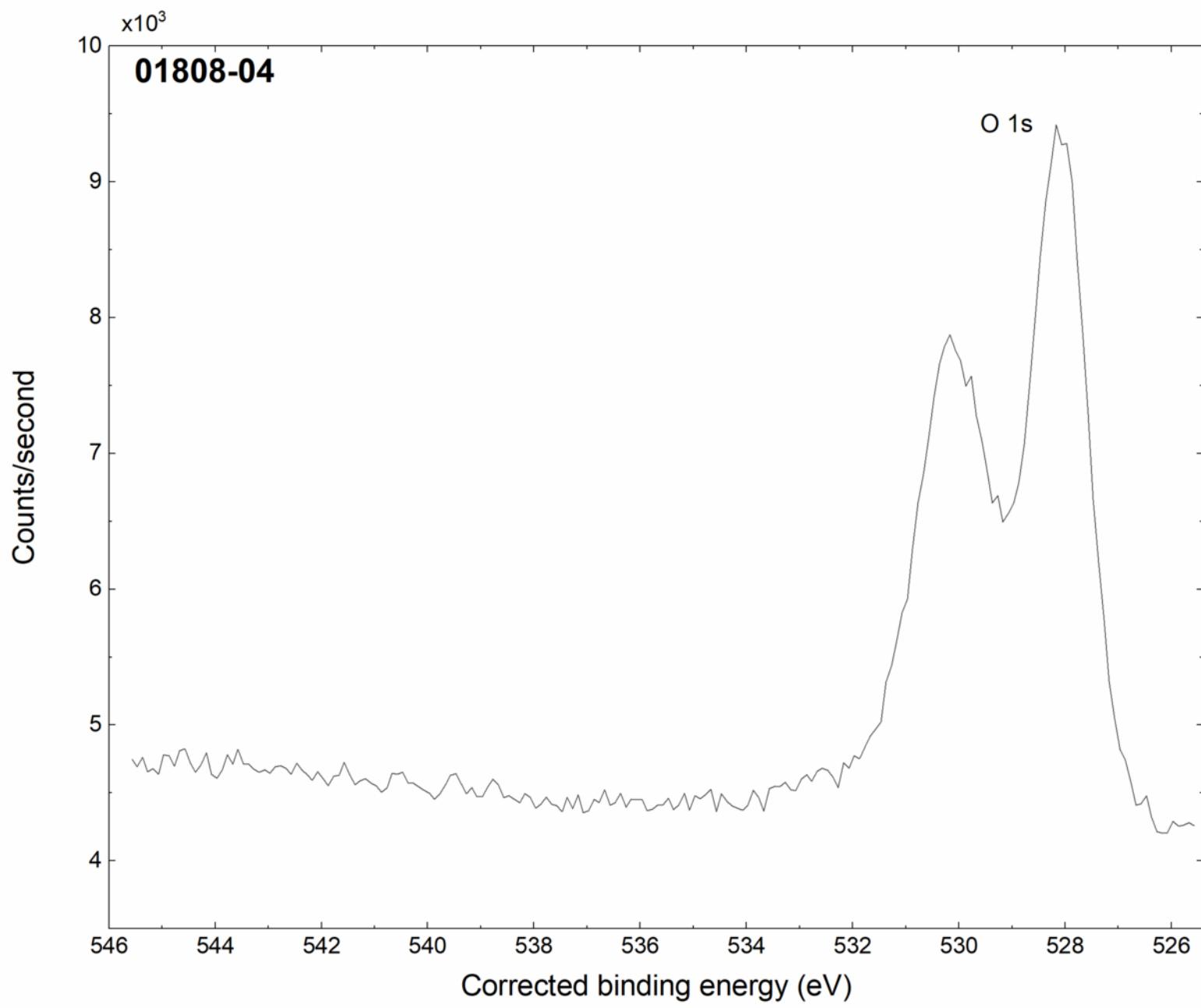


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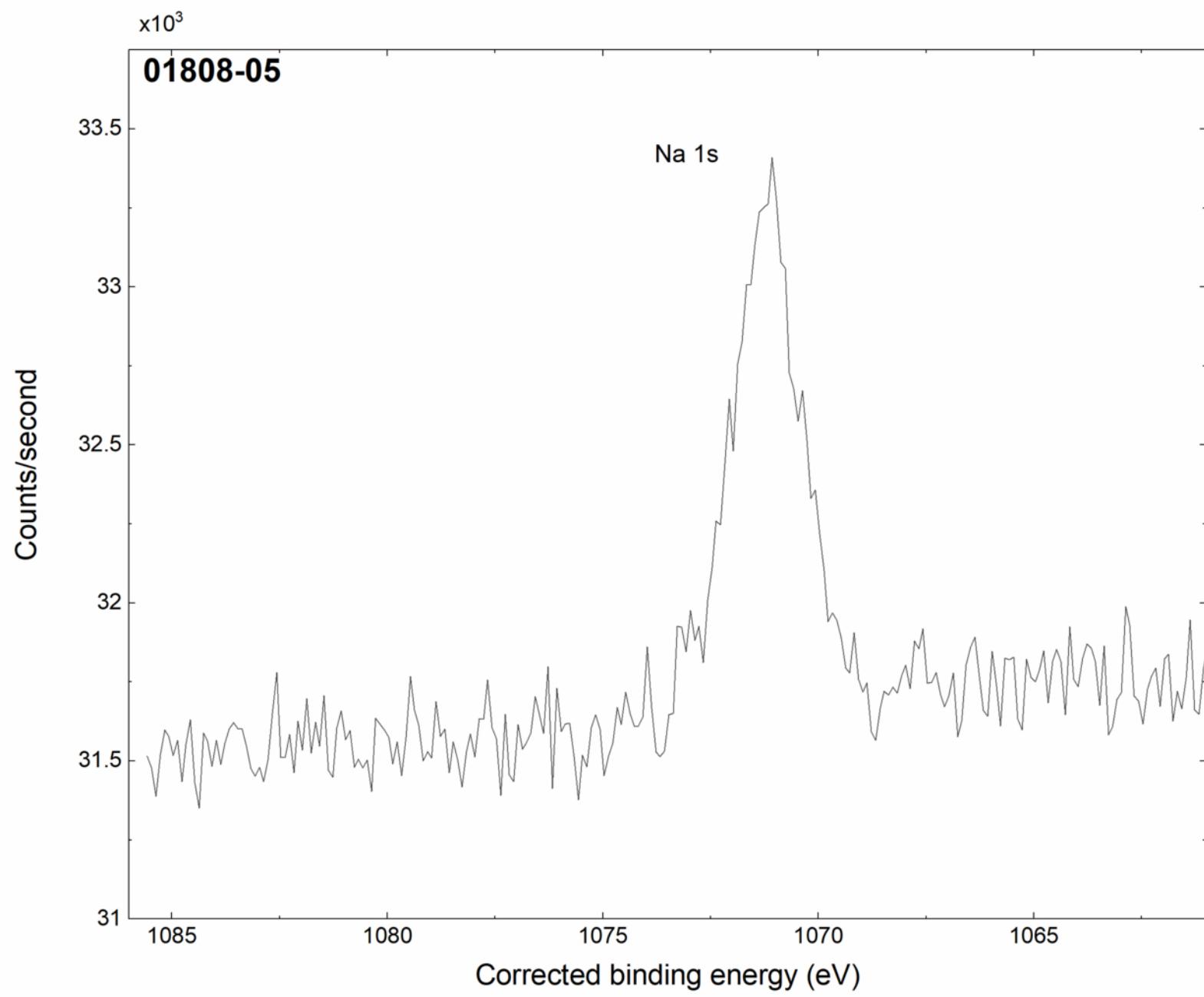


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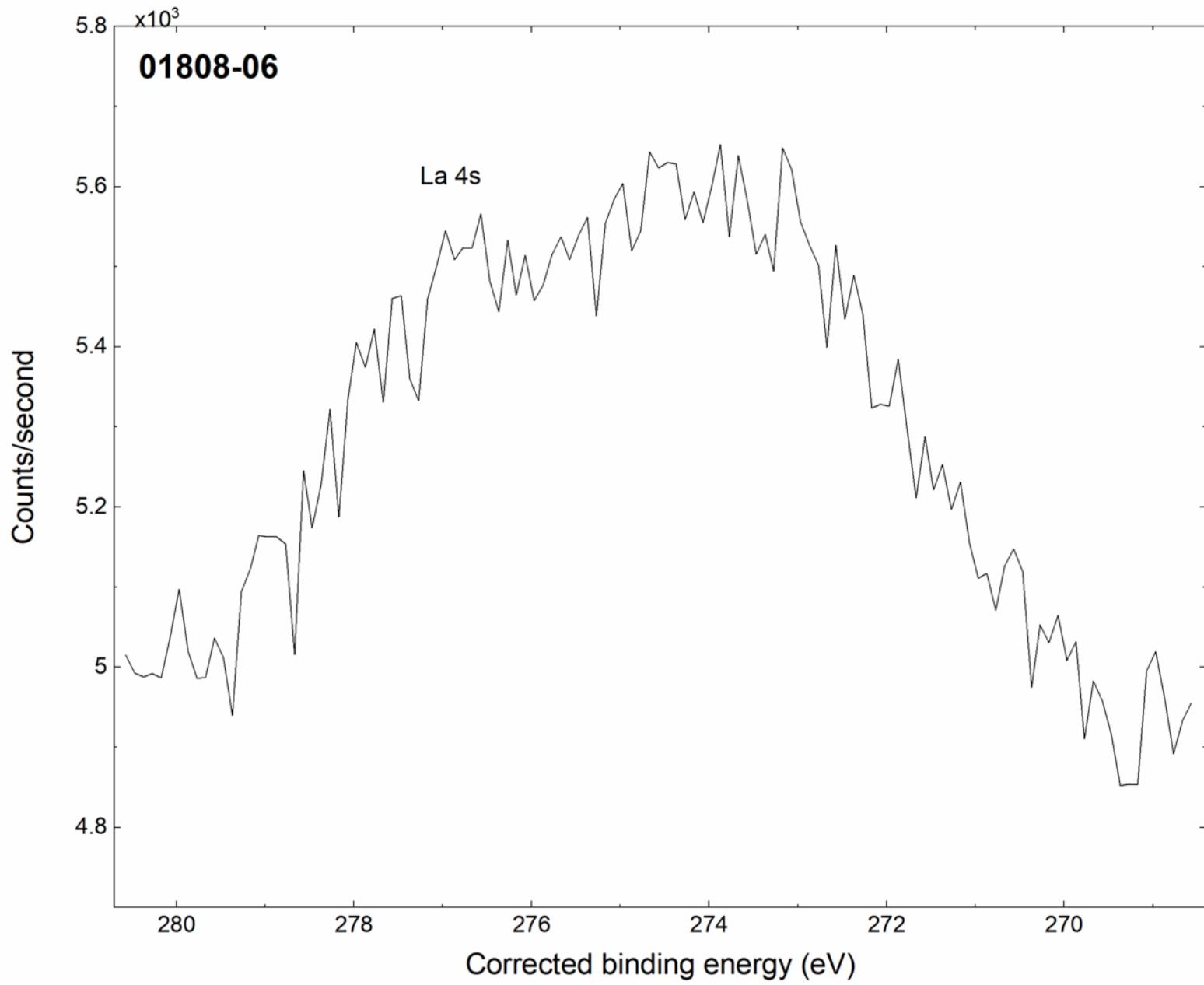


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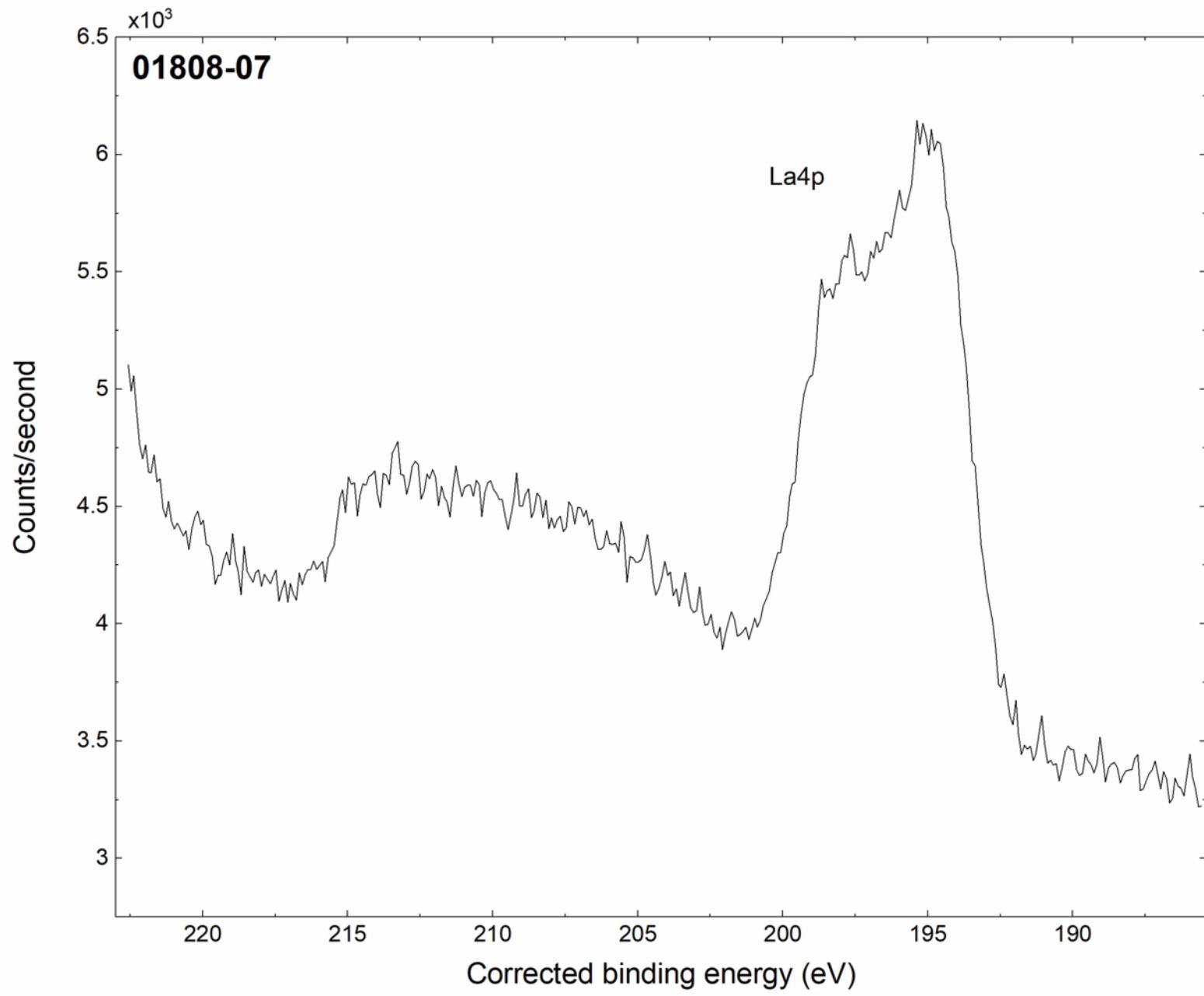




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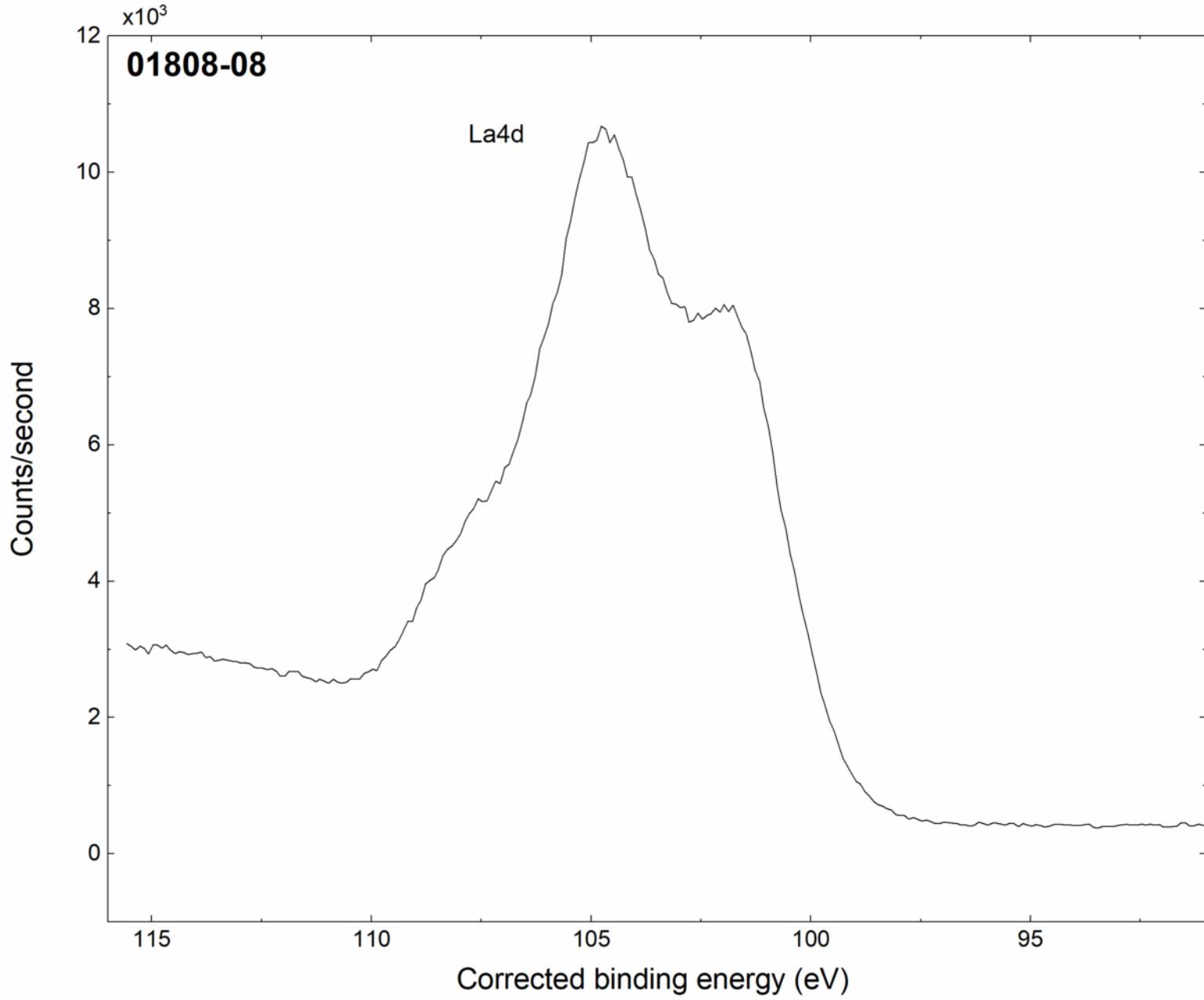


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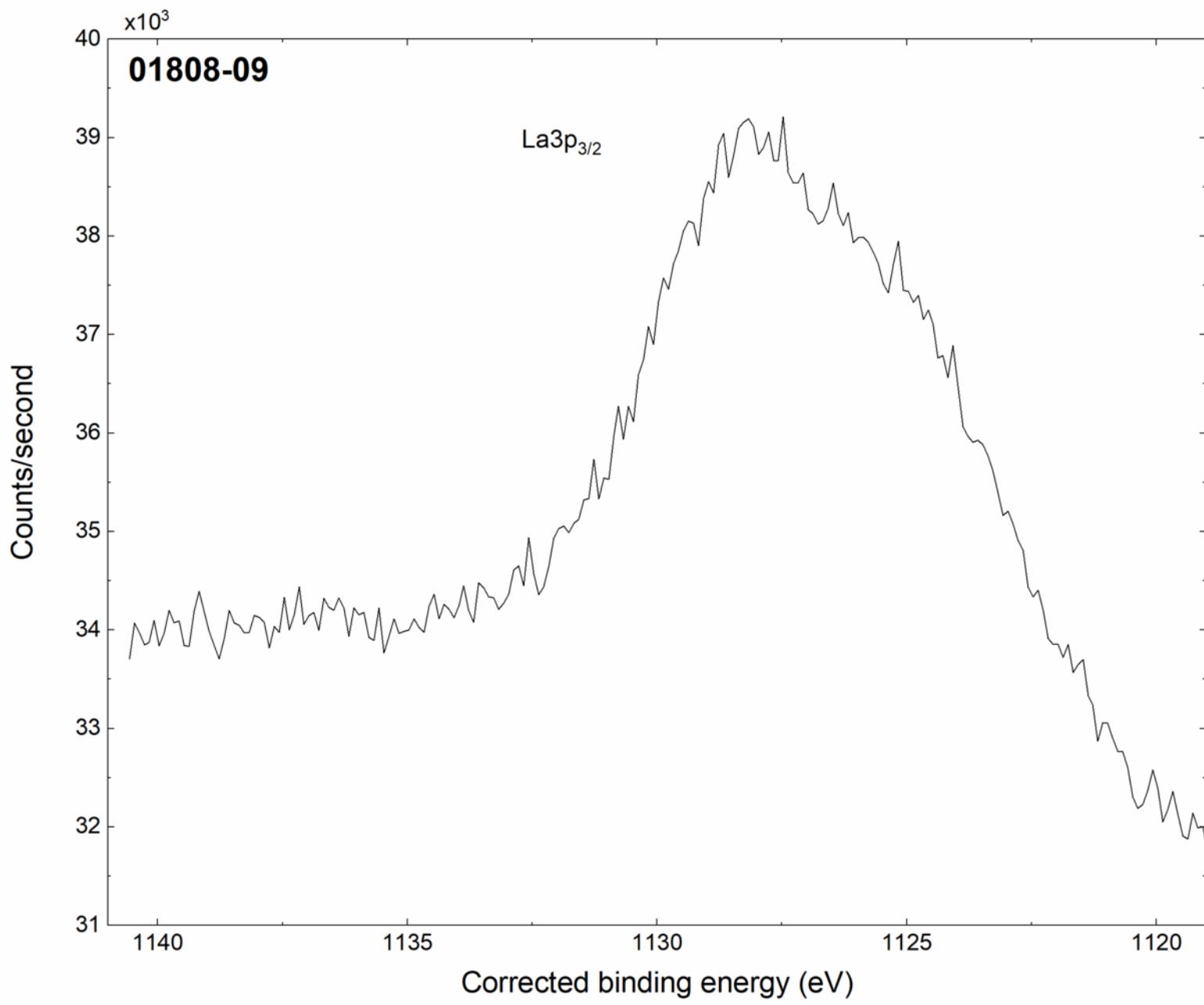


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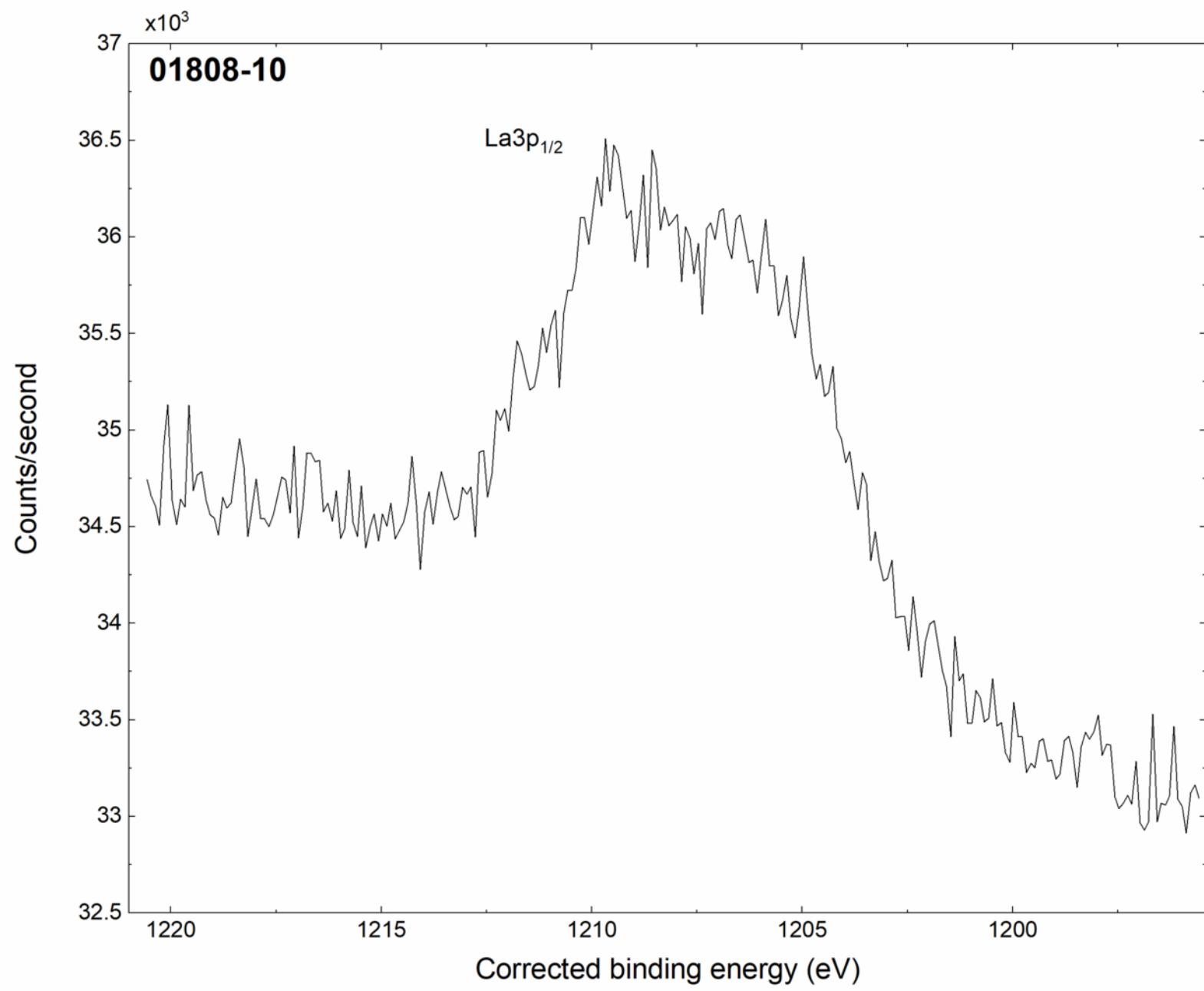


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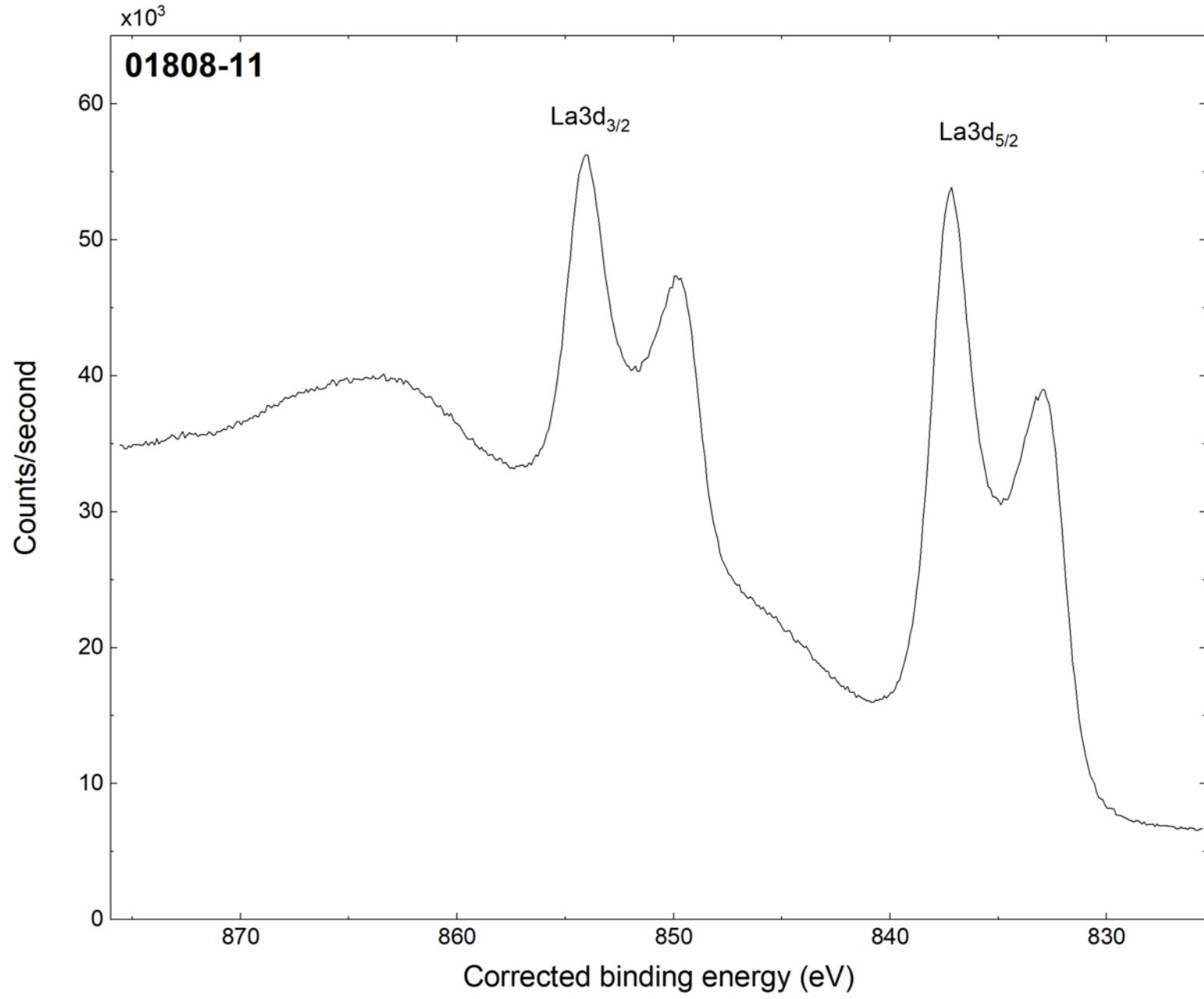


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